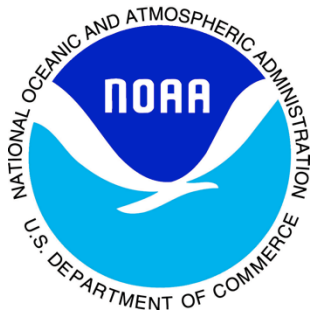


NOAA research to understand and quantify the atmospheric impacts of oil and natural gas operations



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Contributing Scientists:

Ravan Ahmadov, Raul Alvarez, Sergio Alvarez, Arlyn Andrews, **Robert Banta**, **W. Alan Brewer**, **Jerome Brioude**, Steve Brown, Steve Conley, Eric Crosson, Joost de Gouw, Ed Dlugokencky, Bill Dubé, **Peter Edwards**, James Flynn, Felix Geiger, **Jessica Gilman**, Martin Grauss, Mike Hardesty, Andrew Hart, Detlev Helmig, Jack Higgs, Scott Hill, John Holloway, Jacques Hueber, Bryan Johnson, **Anna Karion**, Jon Kofler, Pat Lang, Andy Langford, Barry Lefer, Brian Lerner, Randy Martin, Stuart McKeen, Ben Miller, Steve Monztkka, Don Neff, Tim Newberger, Sam Oltmans, Jeff Peischl, **Gabrielle Pétron**, Bernard Rappenglueck, Chris Rella, **Jim Roberts**, Rachel Russo, Tom Ryerson, Scott Sandberg, **Russ Schnell**, **Christoph Senff**, Barkley Sive, Colm Sweeney, Pieter Tans, Michael Trainer, Tracy Tsai, Bruce Vaughn, Nick Wagner, **Carsten Warneke**, Rebecca Washenfelder, Anne Weickmann, Eric Williams, Dan Wolfe, Sonja Wolter, Cora Young

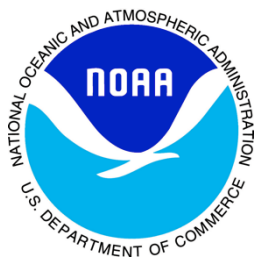
Overview of NOAA research about O&G impacts

NOAA measurements and analysis focus on two O&G basins

- Denver-Julesburg Basin (DJB), Colorado :
 - Monitoring and source sampling since 2008
 - Field intensives in 2008, 2011, 2012
- Uintah Basin, Utah:
 - Intensives in 2012, 2013

Note that the data shown, unless otherwise indicated, are preliminary.

- Please do not cite unless you have received explicit permission of the relevant investigators.



NOAA Research Facilities and Partners

Boulder, Colorado



National Oceanic & Atmospheric Administration (NOAA)

- Earth System Research Laboratory (ESRL), <http://www.esrl.noaa.gov>
 - Chemical Sciences Division (CSD), <http://www.esrl.noaa.gov/csd/>
 - Global Monitoring Division (GMD), <http://www.esrl.noaa.gov/gmd/>
 - Physical Sciences Division (PSD), <http://www.esrl.noaa.gov/psd/>

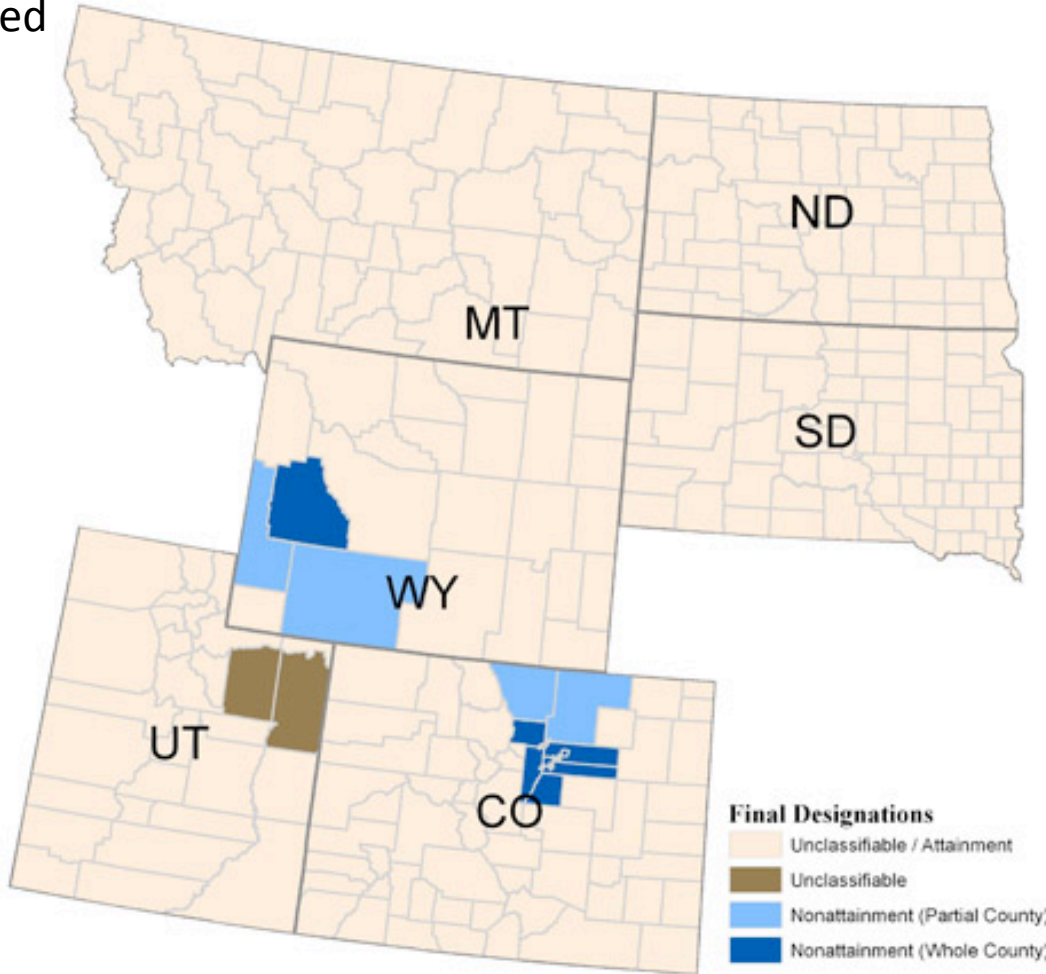
University of Colorado (CU)

- Cooperative Institute for Research in Environmental Sciences (CIRES), <http://cires.colorado.edu/>
- Institute for Arctic and Alpine Research (INSTAAR), <http://instaar.colorado.edu/>

Where is air quality impacted?

EPA Region 8 Final Designations for Surface Ozone

Released
April
2012



Wyoming:

Lincoln, Sublette, Sweetwater are non-attainment in the **winter time**

Colorado:

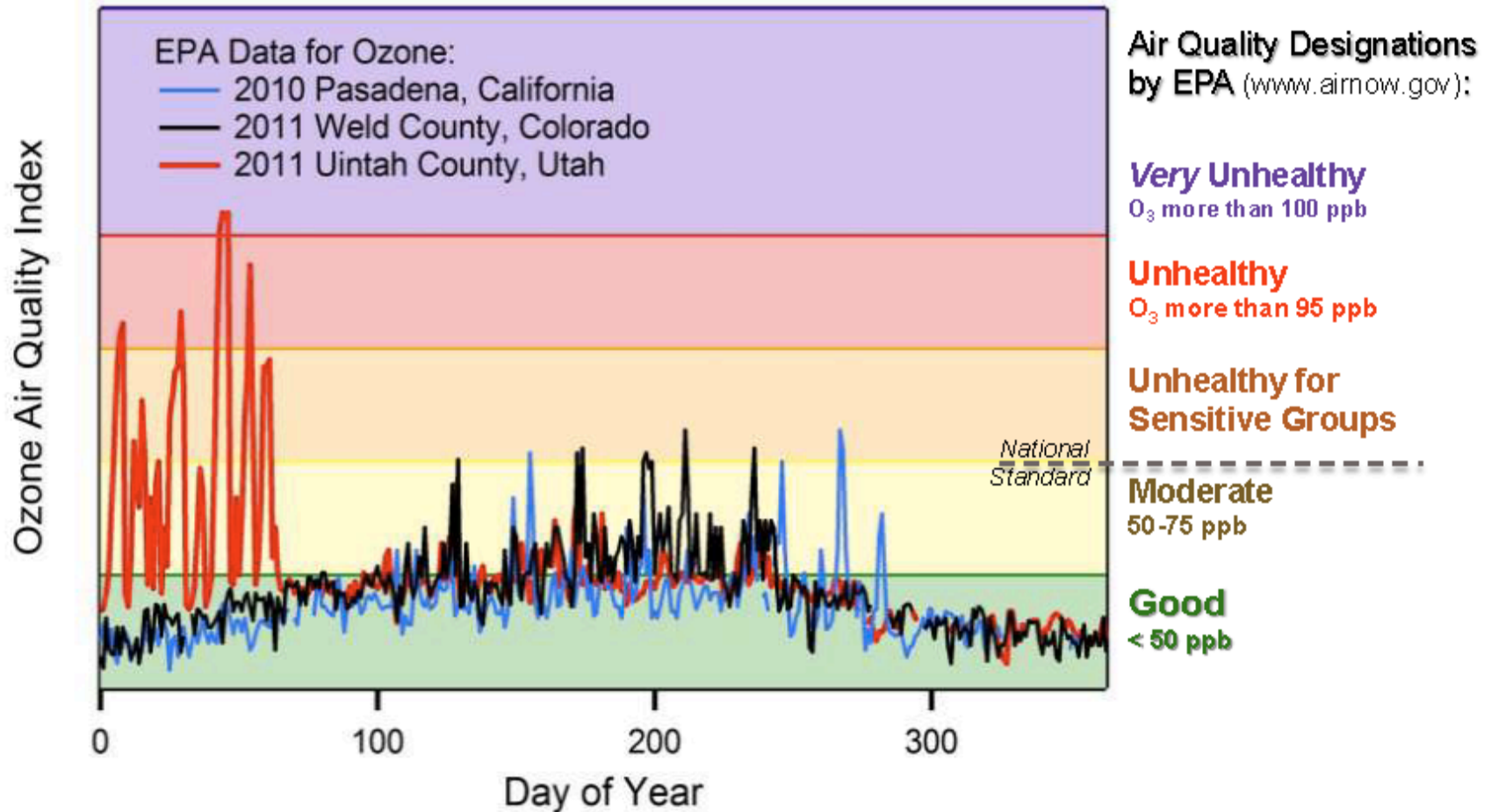
Adams, Arapahoe, Boulder, Broomfield, Denver, Douglas, Jefferson, Larimer, Weld are non-attainment in the **summer time**

Utah:

Uintah and Duchesne unclassifiable

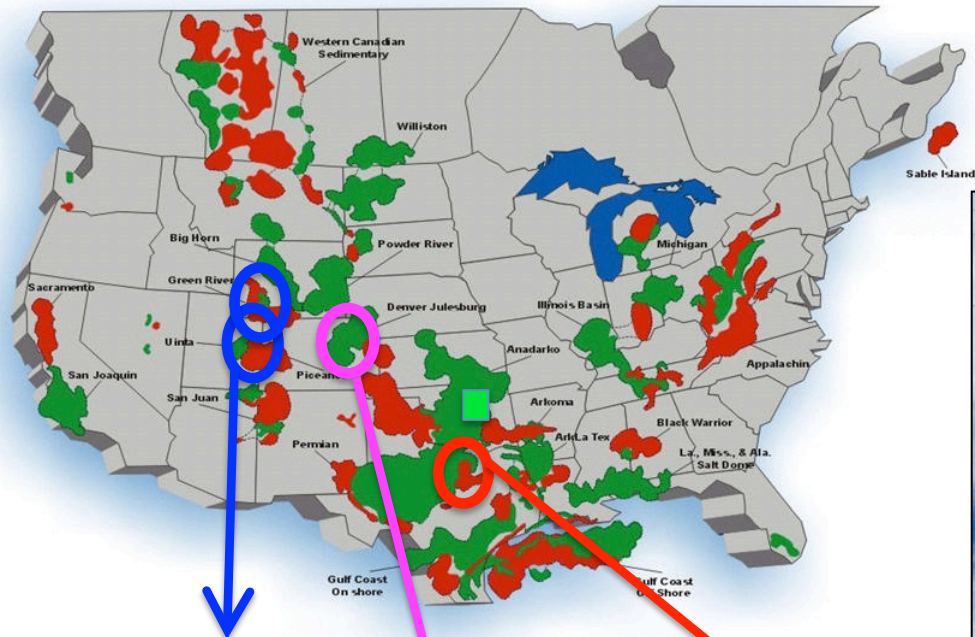
Photochemical formation of ozone (O_3)

Ground-level O_3 degrades air quality



Past and Ongoing NOAA Monitoring & Field Intensives

Map of unconventional **oil** and **gas** reservoirs and dates of NOAA monitoring and campaigns



Wyoming 2008
Utah 2012, 2013

Colorado
2008-present

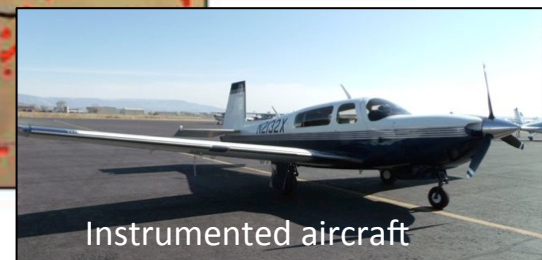
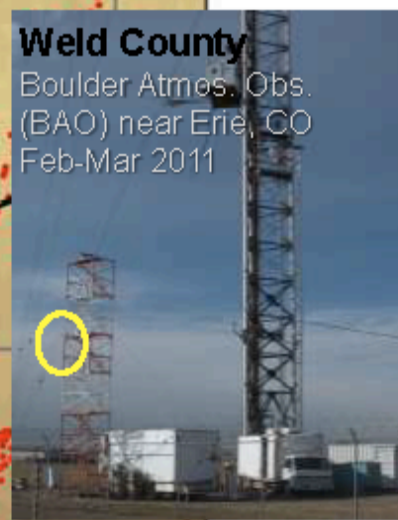
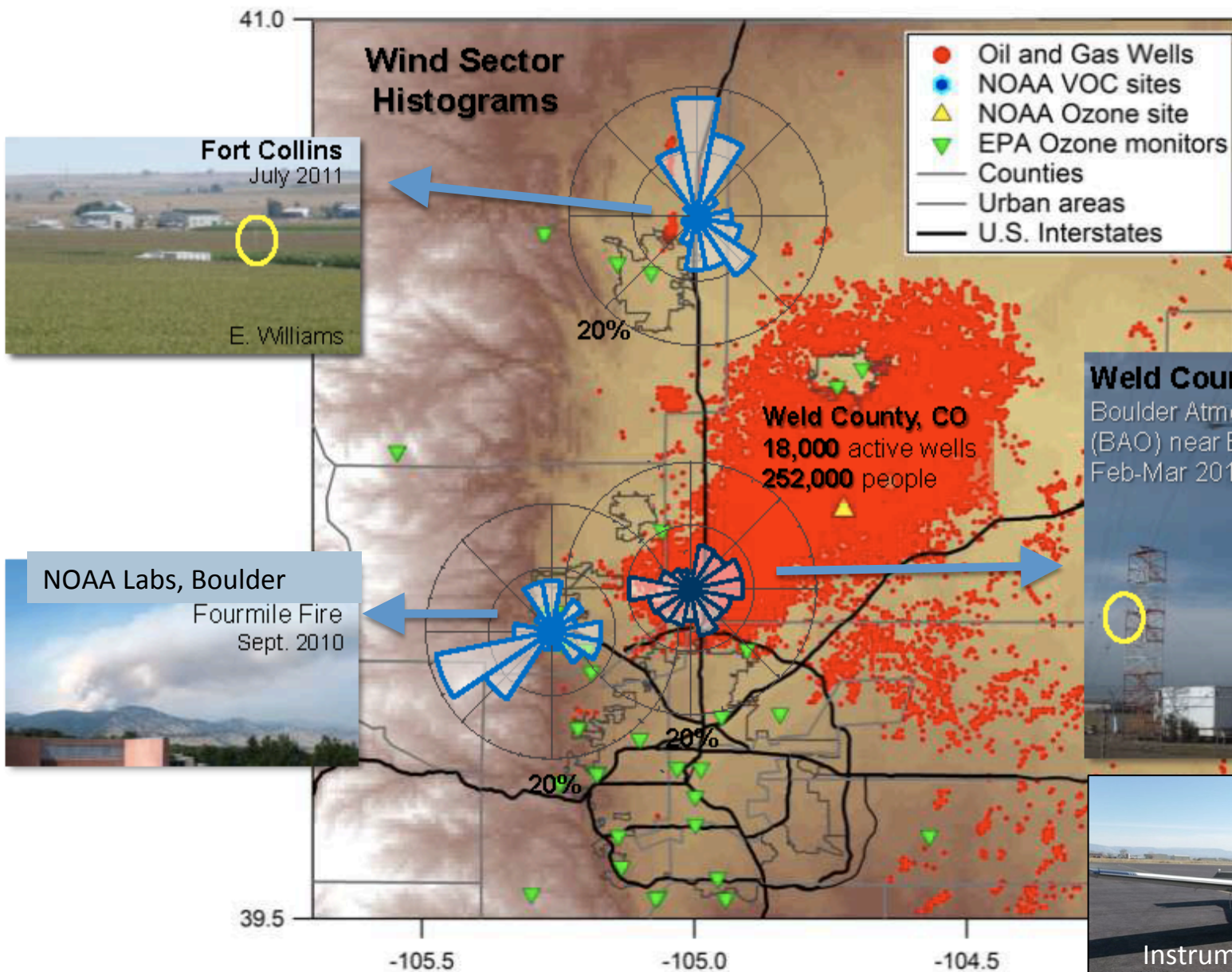
Texas
2013

RESEARCH QUESTIONS:

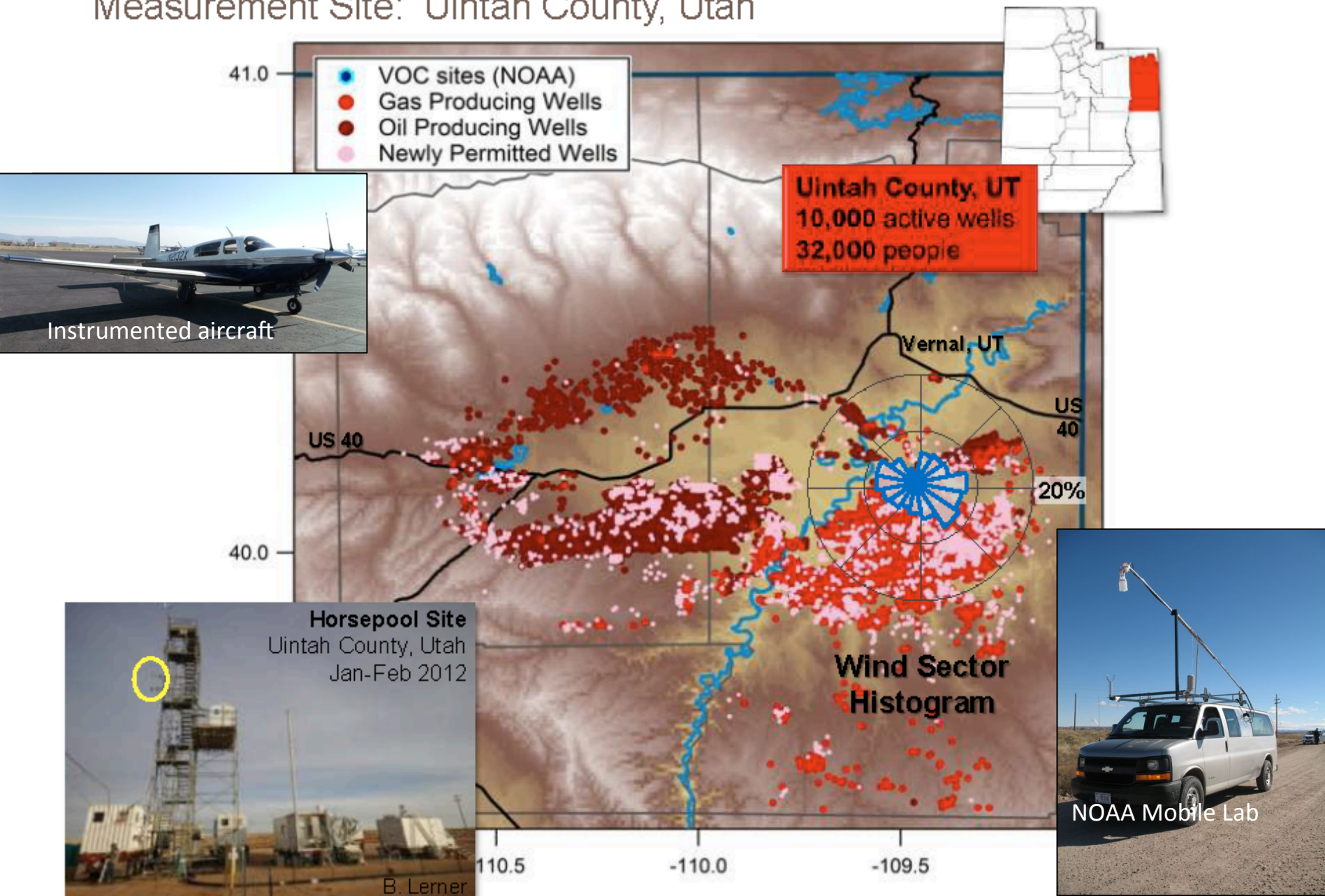
- Emissions from O&G
- Summertime ozone drivers
- Wintertime ozone drivers



Measurement Sites: Colorado Front Range



Measurement Site: Uintah County, Utah



CO/UT NOAA O&G Studies in Literature

44

● Oil and Gas Wells

LETTERS Schnell, et al. (2009) nature geoscience
Rapid photochemical production of ozone at high concentrations in a rural site during winter
 Russell C. Schnell^{1*}, Samuel J. Oltmans², Ryan R. Neely³, Maggie S. Endres², John V. Molnar³ and Allen B. White¹

Source signature of volatile organic compounds (VOCs) from oil and natural gas operations in northeastern Colorado

Jessica B. Gilman, Brian M. Lerner, William C. Kuster, and Joost de Gouw

Environ. Sci. Technol., Just Accepted Manuscript • DOI: 10.1021/es304119a • Publication Date (Web): 14 Jan 2013



Gilman et al. (2013)

ATMOSPHERIC CHEMISTRY
Wyoming winter smog
 Surface ozone levels are expected to be high in polluted regions during summer months. Observations from Wyoming in February 2008 indicate that equally high concentrations of ozone can be produced during winter.
 NATURE GEOSCIENCE | VOL 2 | FEBRUARY 2009 | www.nature.com/naturegeoscience

40

Uintah Basin

Front Range Denver-Julesburg Basin
 July 2008

Atmos. Chem. Phys. Discuss., 13, 7503–7552, 2013
 www.atmos-chem-phys-discuss.net/13/7503/2013/
 doi:10.5194/acpd-13-7503-2013
 © Author(s) 2013. CC Attribution 3.0 License.



This discussion paper is/has been under review for the journal Atmospheric Chemistry and Physics (ACP). Please refer to the corresponding final paper in ACP if available.

Ozone photochemistry in an oil and natural gas extraction region during winter: simulations of a snow-free season in the Uintah Basin, Utah

P. M. Edwards^{1,2}, C. J. Young^{1,*}, K. Aikin^{1,2}, J. A. deGouw^{1,2}, W. P. Dubé^{1,2}, F. Geiger³, J. B. Gilman^{1,2}, D. Helmig⁴, J. S. Holloway^{1,2}, J. Kercher⁵, B. Lerner^{1,2}, R. Martin⁶, R. McLaren⁷, D. D. Parrish¹, J. Peischl^{1,2}, J. M. Roberts¹, T. B. Ryerson¹, J. Thornton⁸, C. Warneke^{1,2}, E. J. Williams¹, and S. S. Brown¹

Hydrocarbon emissions characterization in the Colorado Front Range: A pilot study Pétron, et al. (2012)

Gabrielle Pétron^{1,2}, Gregory Frost^{1,2}, Benjamin R. Miller^{1,2}, Adam I. Hirsch^{1,3}, Stephen A. Montzka², Anna Karion^{1,2}, Michael Trainer², Colm Sweeney^{1,2}, Arlyn E. Andrews², Lloyd Miller⁴, Jonathan Kolfer^{1,2}, Amnon Bar-Ilan⁵, Ed J. Dlugokencky², Laura Patrick^{1,2}, Charles T. Moore Jr.⁶, Thomas B. Ryerson², Carolina Siso^{1,2}, William Kolodzey⁷, Patricia M. Lang², Thomas Conway², Paul Novelli², Kenneth Masarie², Bradley Hall², Douglas Guenther^{1,2}, Duane Kitzis^{1,2}, John Miller^{1,2}, David Welsh², Dan Wolfe², William Neff², and Pieter Tans²

JOURNAL OF GEOPHYSICAL RESEARCH, VOL. 117, D04304, doi:10.1029/2011JD016360, 2012

Air sampling reveals high emissions from gas field

Methane leaks during production may offset climate benefits of natural gas.

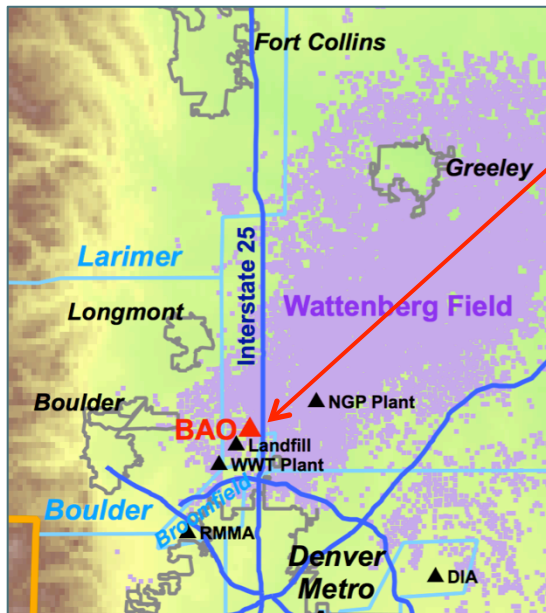
-114 -112 -110 -108 -106 -104 -102

NOAA measurements and analysis of emissions in O&G regions

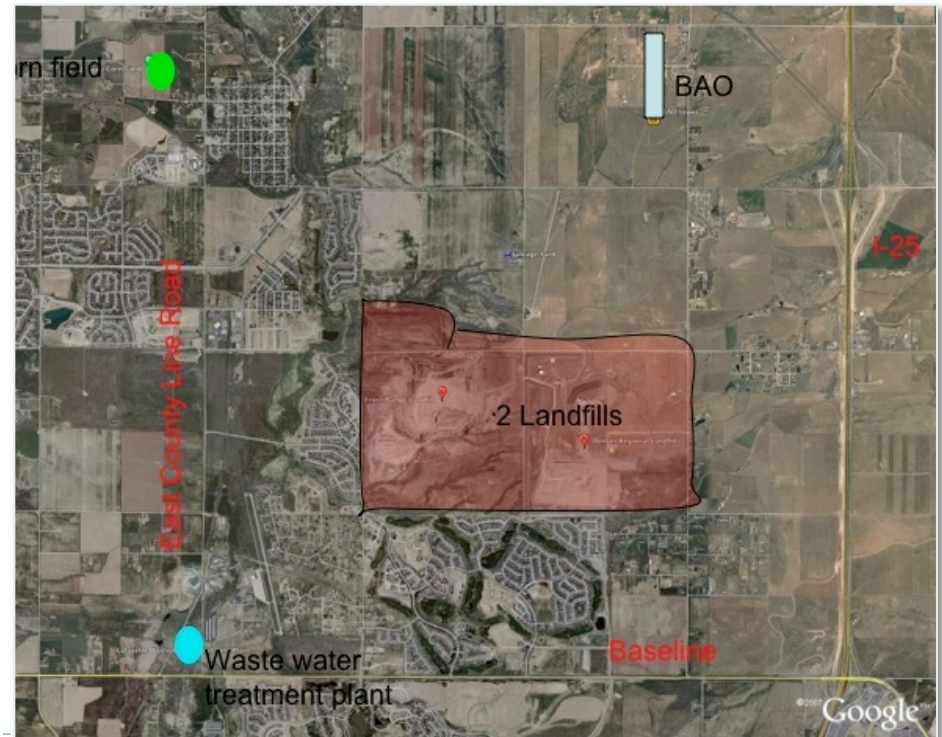
Some examples of top-down emissions analysis

- Top-down = based on atmospheric observations
 - Use coincident measurements of multiple species
 - Regional emissions quantification
 - Source identification and characterization
1. Tall tower + mobile sampling, DJB summer 2008 (Pétron et al., 2012)
 2. Surface sampling, DJB winter 2011/ Uintah winter 2012 (Gilman et al., 2013)
 3. Aircraft sampling, Uintah winter 2012/ DJB spring 2012 (Karion, Sweeney, et al.)
 4. Mobile sampling, Uintah winter 2012 (Pétron, Geiger/ Warneke, Edwards, et al.)

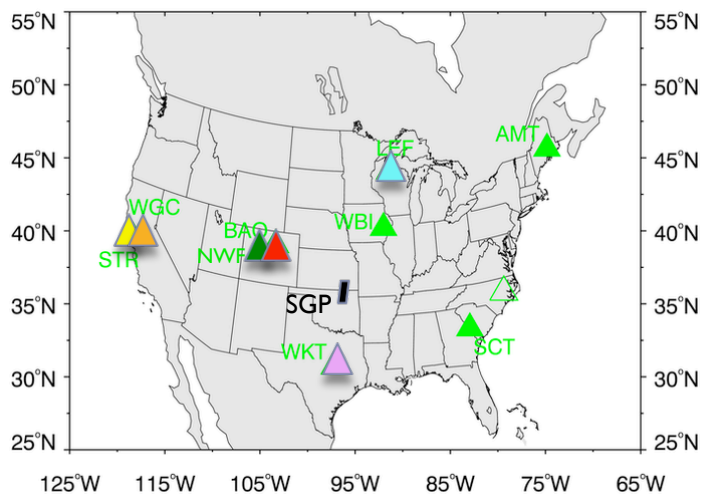
Boulder Atmospheric Observatory



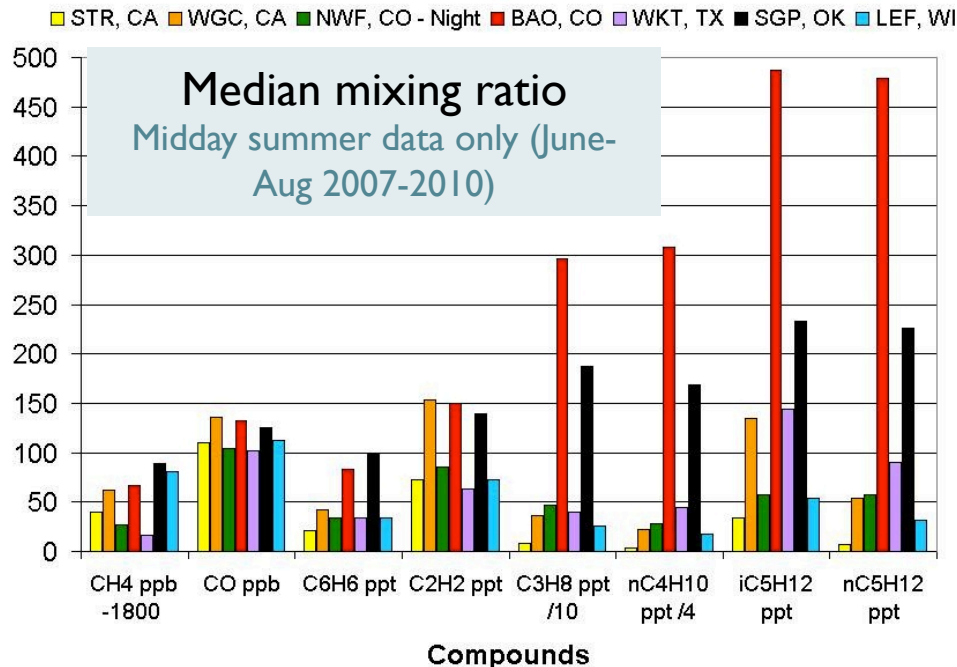
- 300 meter tall tower
- located in Erie, Weld County
- Instrumented with LICOR (CO₂) and TECO (CO) in April 2007: sampling from 3 intake heights (22m, 100m, 300m)
- 30 sec- Met Data at three levels
- Equipped to collect discrete air samples from 300 meter level in August 2007. Analyses performed in NOAA Boulder lab.



BAO: Distinct Alkane Signature



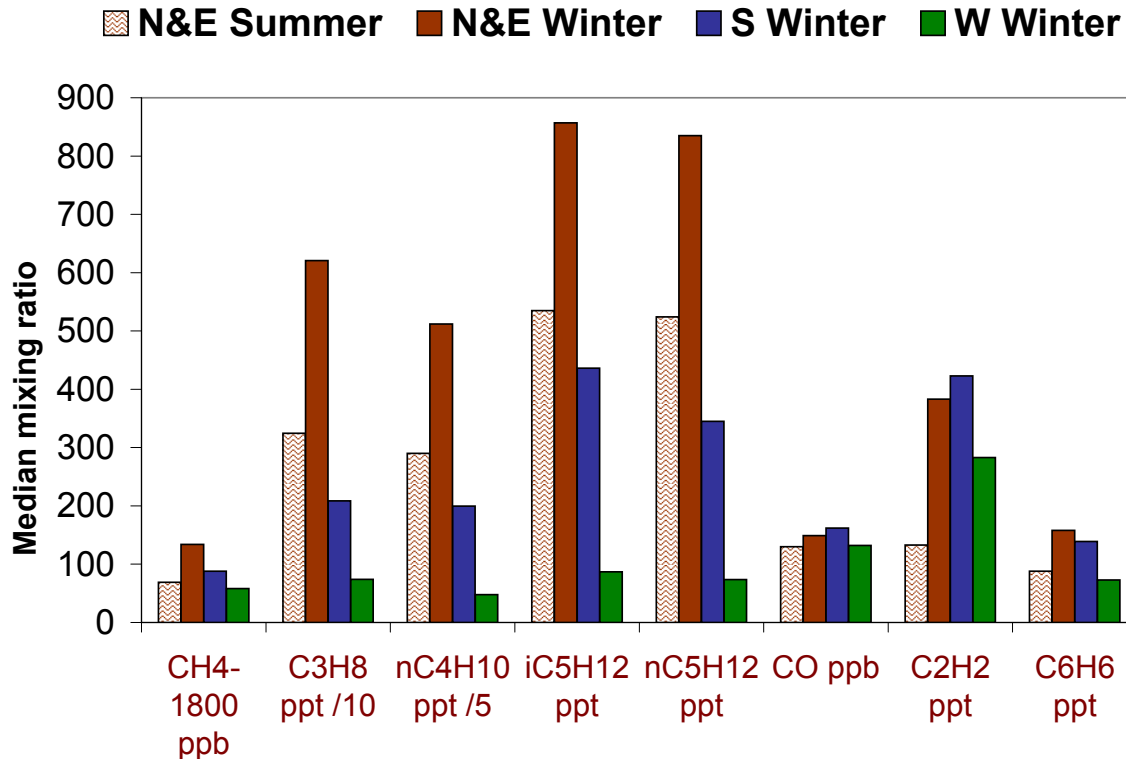
NOAA Tall Tower Measurement and Sampling Network (PI Arlyn Andrews)



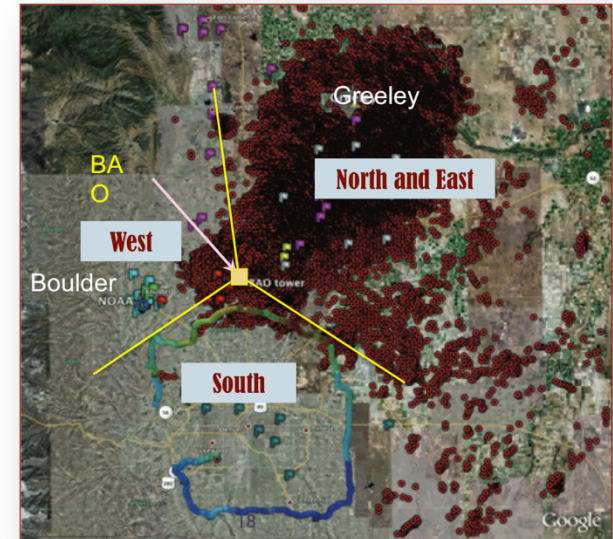
Air samples collected at the BAO and SGP* have a strong alkane signature.

* SGP is a NOAA aircraft site in Northern Oklahoma. Samples collected below 650 meters were used for this analysis.

BAO: Data Filtered By Wind Sector



Strongest alkane signature in North & East wind sector



Midday Data from the BAO (August 2007-April 2010).

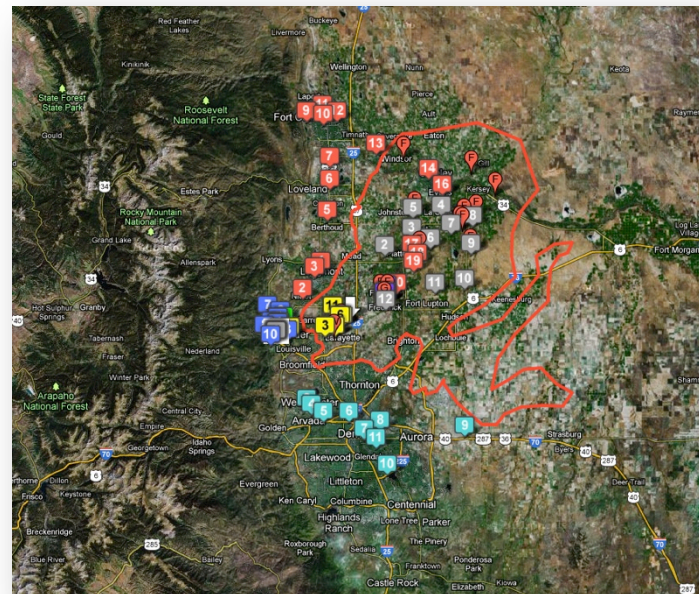
Wind sector designation based on 30-min average (prior to sample collection) wind direction and wind speed (data retained if $|w.speed| > 2.5$ m/s).

Mobile lab sampling of methane source chemical signatures in the Front Range

- ▶ Mobile Platform to sample close to sources
- ▶ High-frequency stable analyzers to detect plumes and target flask sampling
- ▶ Discrete air sampling for multi-species chemical analyses in the NOAA lab

Toyota Prius equipped with:

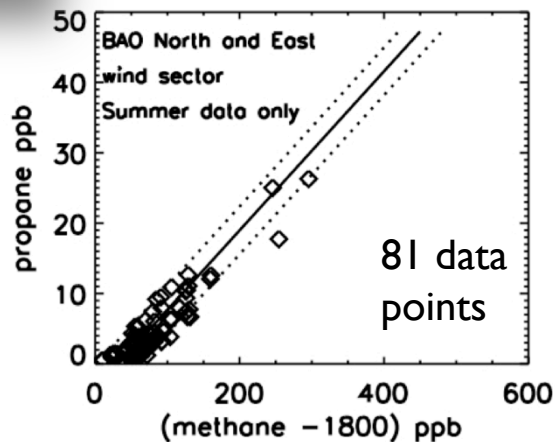
- Fast response CO₂ and CH₄ analyzer (Picarro)
- Real Time Display of Measurements
- GPS
- Programmable Flask Package (PFP with 12 sampling glass flasks) and Programmable Compressor Package (PCP) with GPS



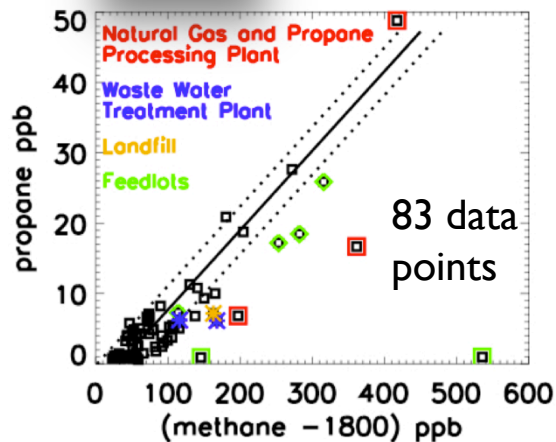
Very Strong Correlations between Alkanes



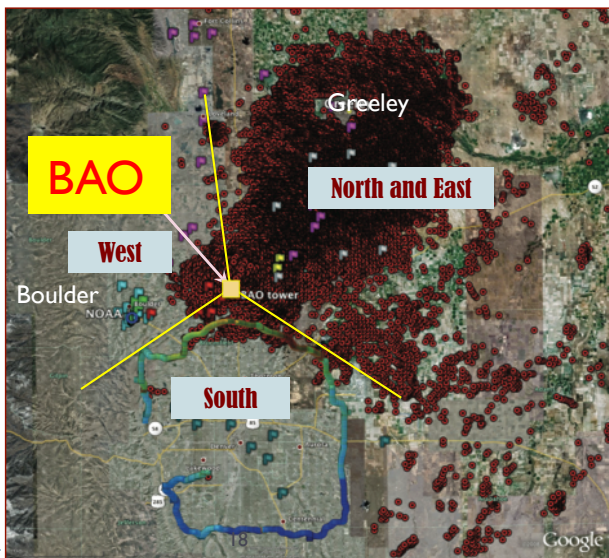
BAO N&E wind sector



Mobile Lab



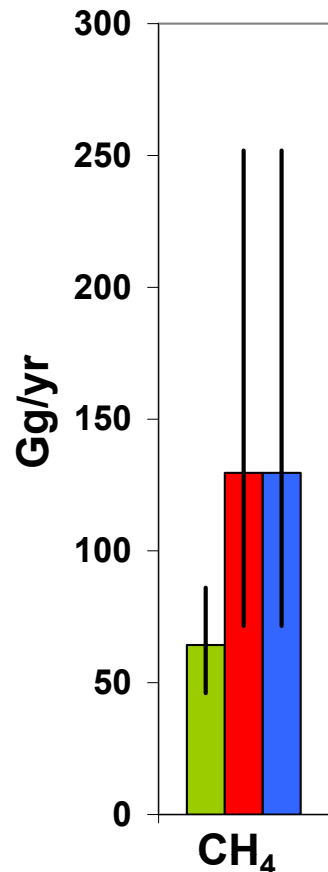
- Methane is strongly correlated with propane.
- Samples collected close to feedlots, a landfill, a waste water treatment plant have enhanced methane compared to the other samples.



We use the measured atmospheric propane-to-methane enhancement ratios observed at the BAO tall tower and at the surface across the Front Range to evaluate the proportion of flashing and venting emissions.

Top-down Estimates versus Inventory

Bottom-up Emissions
 Top-Down BAO/
 Mobile Lab Emissions

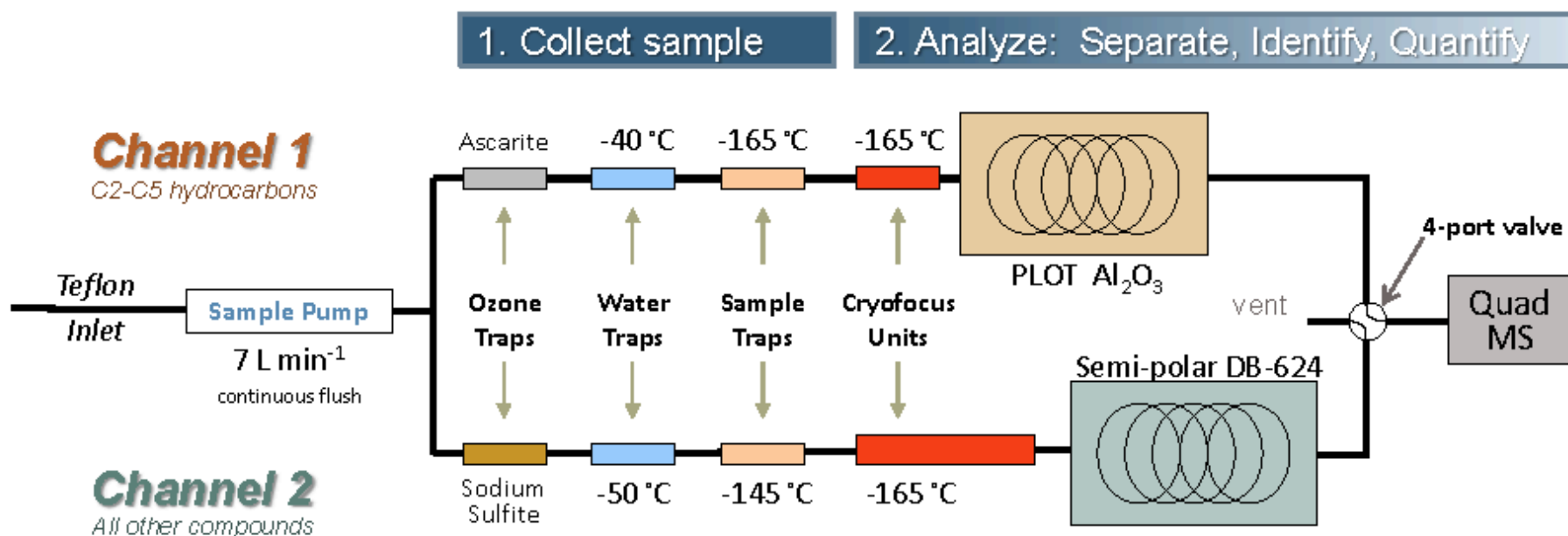


Constraints	Bottom-up inventory	Enhancement Ratio Method
Flash emissions for tanks (total VOC mass) in O ₃ NAA	X	X
Fugitive emissions estimates (volume of raw gas)	X	
Raw gas composition profile (average or subset)	X	X
Flash emissions composition profile	X	X
Atmospheric enhancement ratios		X

- Fugitives emissions of CH₄ are likely underestimated in bottom-up inventory for 2008.
- Still very large uncertainties on top-down estimates.
- **We need truly independent methods to evaluate inventories!**

VOC instrumentation

Gas chromatograph-Mass Spectrometer (GC-MS)



- **5 min sample collection**
 - 350 mL sample for each channel
- **25 min sample analysis**
- **60+ species analyzed**
 - C2 to C11 hydrocarbons
 - C6-C9 aromatics, OVOCs, etc.

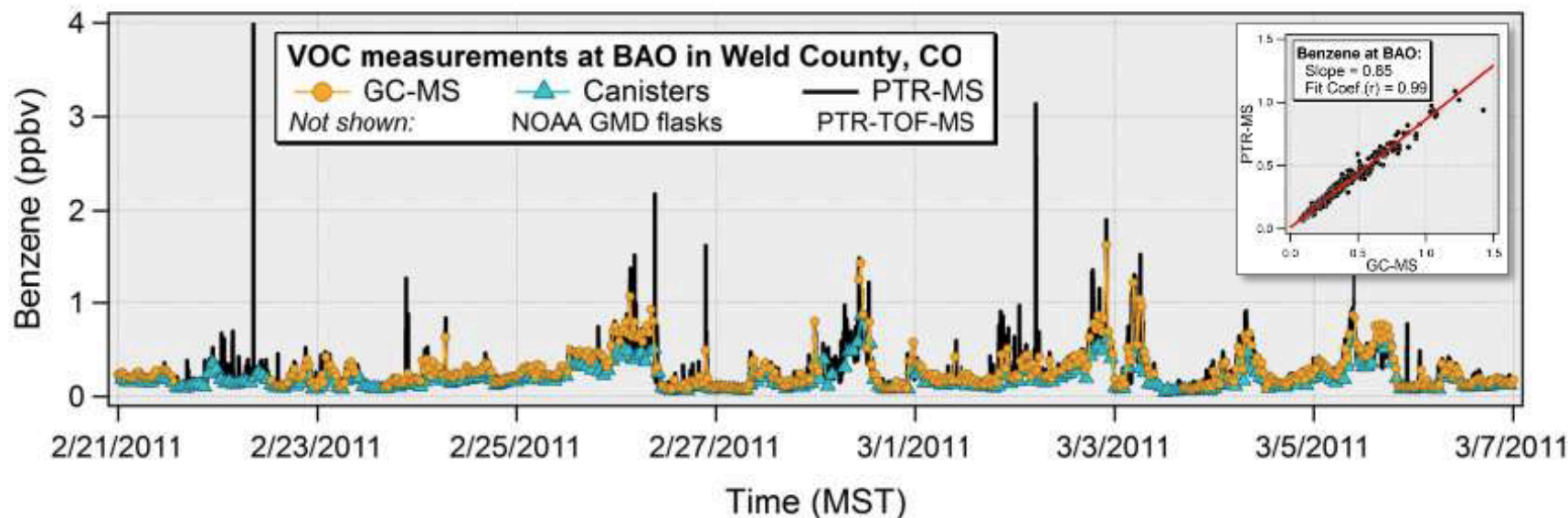
GC-MS Instrument Specs*:

Limit of detection	< 1 - 10 ppt
Precision	0.5% to 5%
Accuracy	15 to 25%

* values are compound dependent

VOC instrumentation in Colorado study

Summary of all measurement techniques



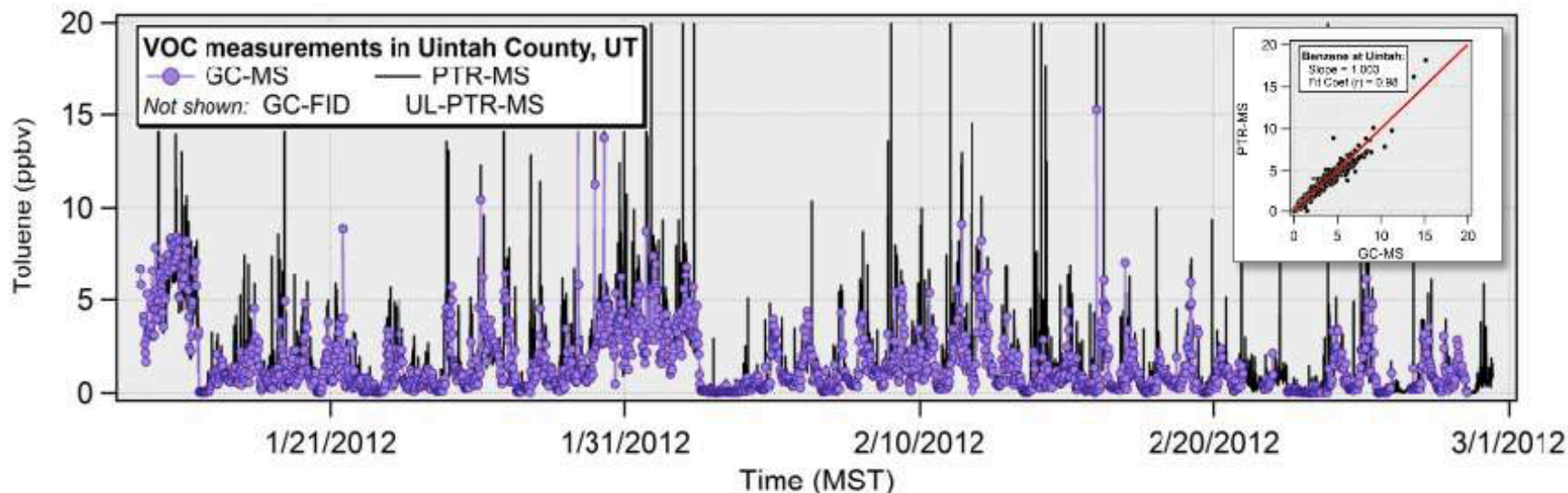
Discrete measurements: In-situ Gas Chromatograph-Mass Spectrometer (GC-MS) *NOAA CSD*
Whole air canister samples analyzed by GC-MS *Appalachian St. Univ.*

Continuous measurements: Proton Transfer Reaction-Mass Spectrometer (PTR-MS) *NOAA CSD*
Proton Transfer Reaction-Time of Flight (TOF)-Mass Spec. *NCAR*

Long-term measurements: Programmable Flask Package (PFP) at 300m at BAO *NOAA GMD*

VOC instrumentation in Utah Study

Summary of all measurement techniques



Discrete measurements:	In-situ Gas Chromatograph-Mass Spectrometer (GC-MS) Whole air canister samples analyzed by GC-MS	NOAA CSD Appalachian St. Univ.
	In-situ GC-FID and Total Non-Methane Hydrocarbons Programmable Flask Package (PFP) in mobile lab	Utah State Univ. NOAA GMD
Continuous measurements:	Proton Transfer Reaction-Mass Spectrometer (PTR-MS) Proton Transfer Reaction-Time of Flight (TOF)-Mass Spec.	NOAA CSD NCAR
	Ultra Light Weight PTR-MS in mobile lab	Karlsruhe Inst. Tech.
Long-term measurements:	Programmable Flask Package (PFP) at 300m at BAO	NOAA GMD

Emission Source Comparison

Gasoline combustion vs. Natural gas production

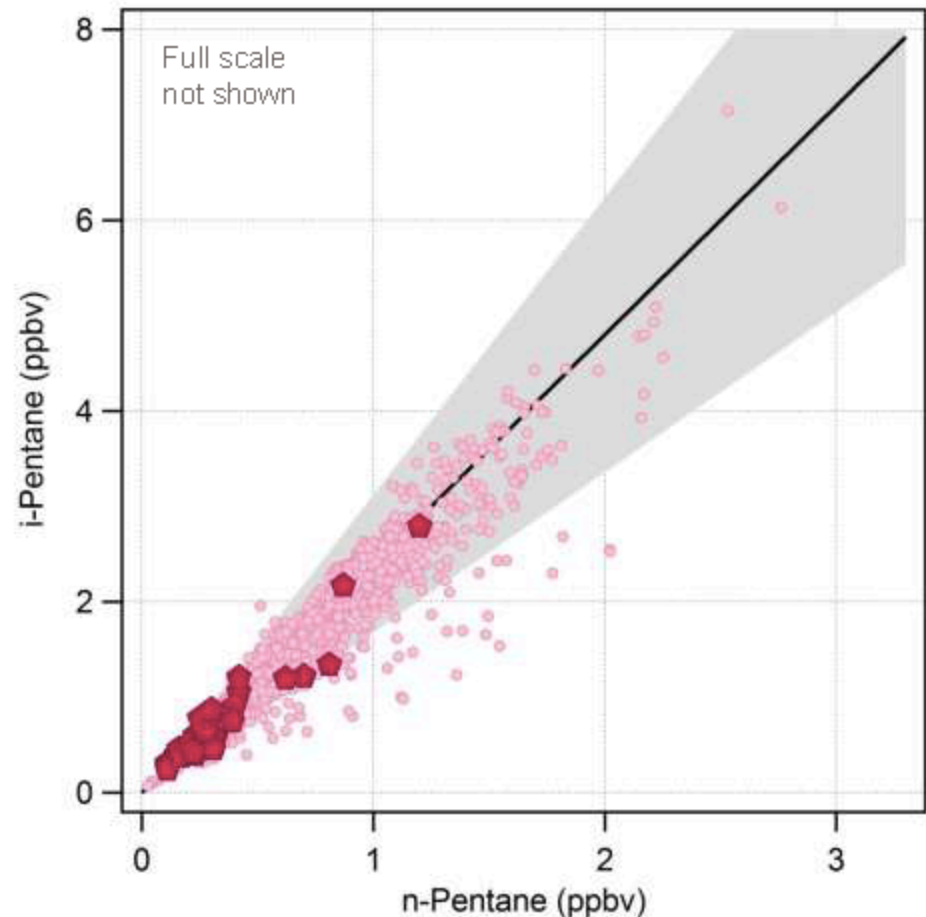
Enhancement Ratios (ER)

i-Pentane vs. n-Pentane:

- Urban ER = $2.4 \pm 20\%$
- Pasadena, CA 2010
- 25 US Cities (Baker, 2008)

Enhancement Ratios

- ER is determined from slope of 2-sided linear regression
- Can be used to:
 - Identify emission sources
 - Examine oxidation chemistry
- Ratios minimize the effects of air mass mixing and dilution



Emission Source Comparison

Gasoline combustion vs. Natural gas production

Enhancement Ratios (ER)

i-Pentane vs. n-Pentane:

- **Urban ER = $2.4 \pm 20\%$**
- Pasadena, CA 2010
- 25 US Cities (Baker, 2008)
- **Natural Gas ER = $0.88 \pm 20\%$**
- ▲ Uintah Co., UT 2012
- ▼ Weld Co., CO 2011

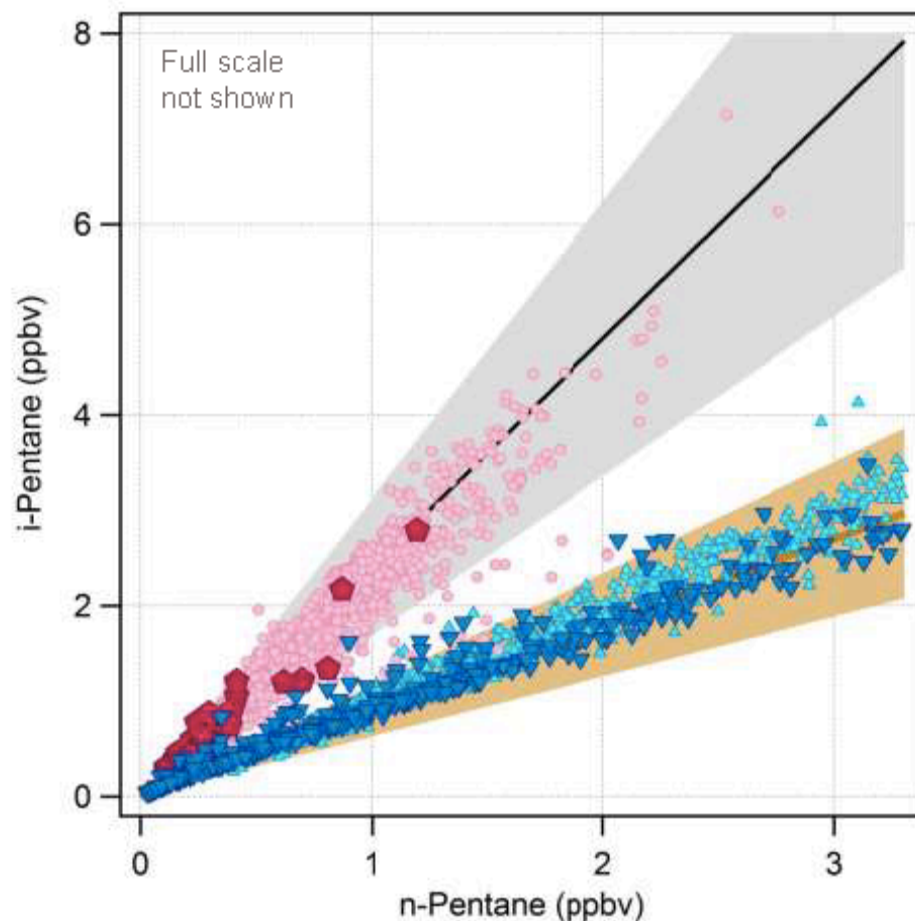
i-Pentane is the most abundant hydrocarbon in gasoline

Source	iso- to n-Pentane Ratio*
Liquid gasoline	2.9
Tailpipe emissions	2.6

*Literature values from:

Schauer, et al. (2002) *Environ. Sci. Tech.*

Gentner, et al. (2009) *Environ. Sci. Tech.*



Emission Source Comparison

Gasoline combustion vs. Natural gas production

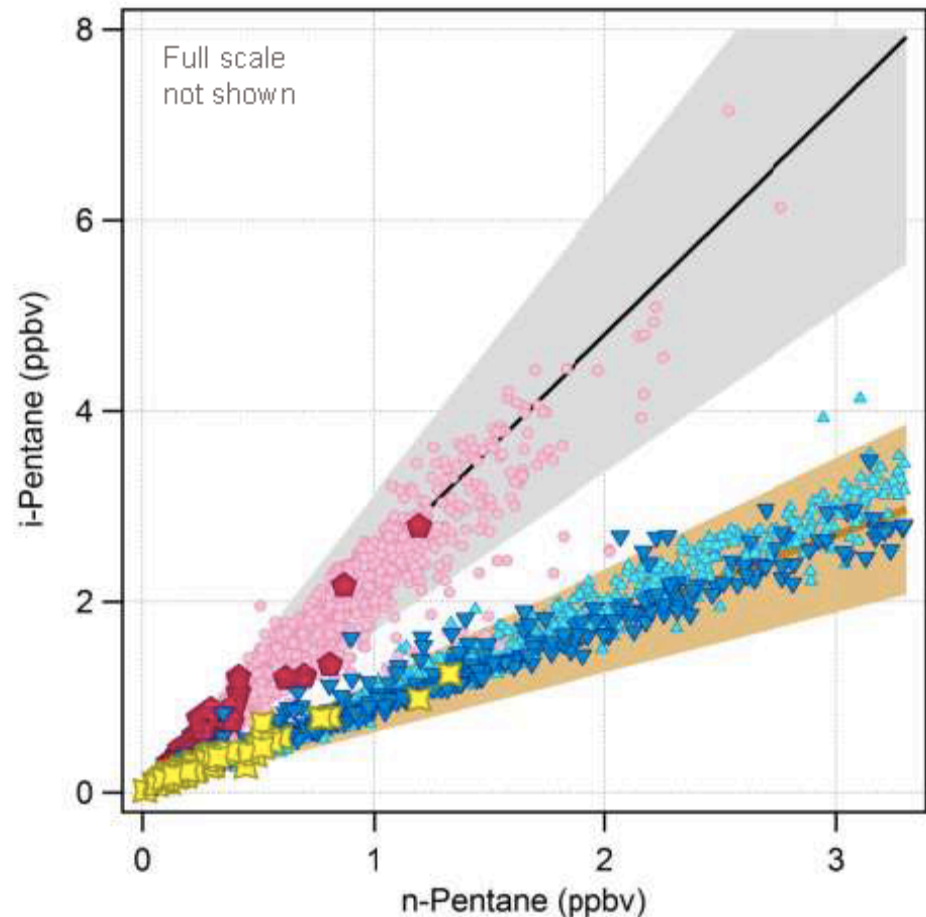
Enhancement Ratios (ER)

i-Pentane vs. n-Pentane:

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- Pasadena, CA 2010
- ◆ 25 US Cities (Baker, 2008)
- **Natural Gas ER = $0.88 \pm 20\%$**
- ▲ Uintah Co., UT 2012
- ▼ Weld Co., CO 2011
- Raw Natural Gas (COGCC rep.)

Raw natural gas composition:

- Report for Colorado Oil and Gas Conservation Commission (COGCC)
- "Greater Wattenberg Area (GWA) Baseline Study" by LT Environmental
 - 77 natural gas wells
 - 6 different production zones
- Values for natural gas samples are in plotted in mole percent ($\text{mol/mol} \times 100\%$)
- $1 \text{ ppbv} = \text{nmol/mol} = \text{mol/mol} \times 10^{-9}$



Emission Source Comparison

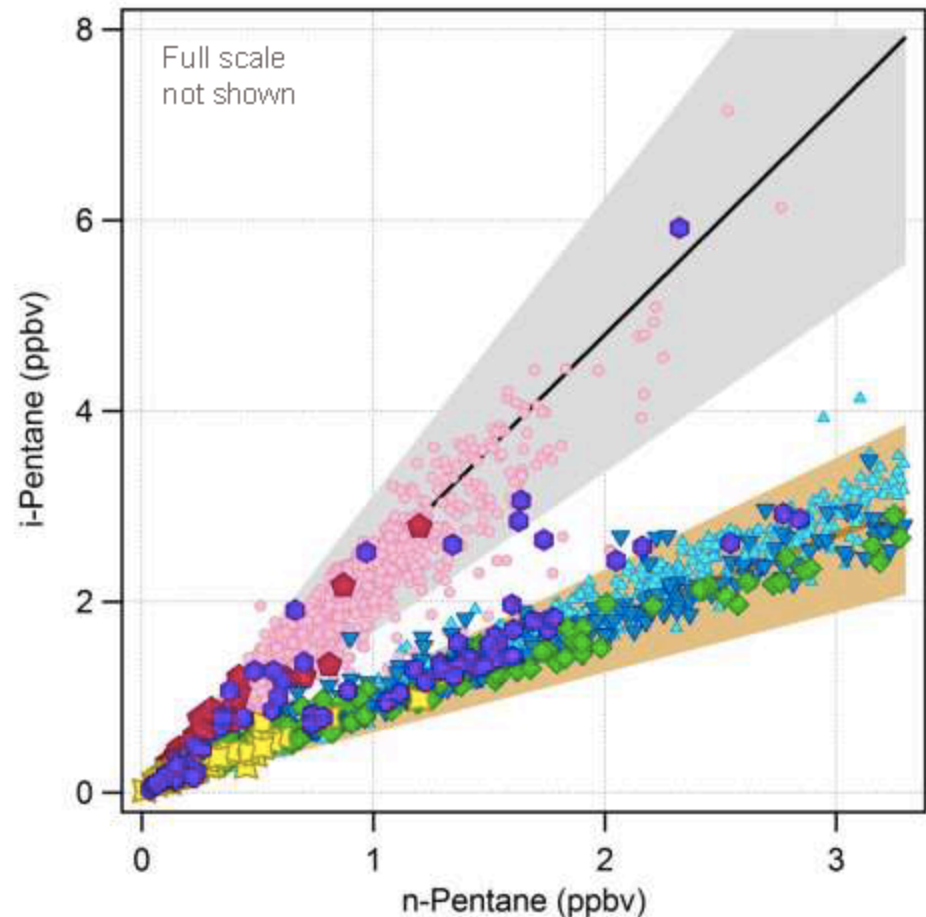
Gasoline combustion vs. Natural gas production

Enhancement Ratios (ER)

i-Pentane vs. n-Pentane:

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 - 25 US Cities (Baker, 2008)
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 - ▲ Uintah Co., UT 2012
 - ▼ Weld Co., CO 2011
 - Raw Natural Gas (COGCC rep.)
- ◆ Fort Collins, CO 2011
- Boulder, CO 2010

All of the measurement sites in Colorado's Front Range were influenced by emissions from oil and gas operations in the area



Emission Source Comparison

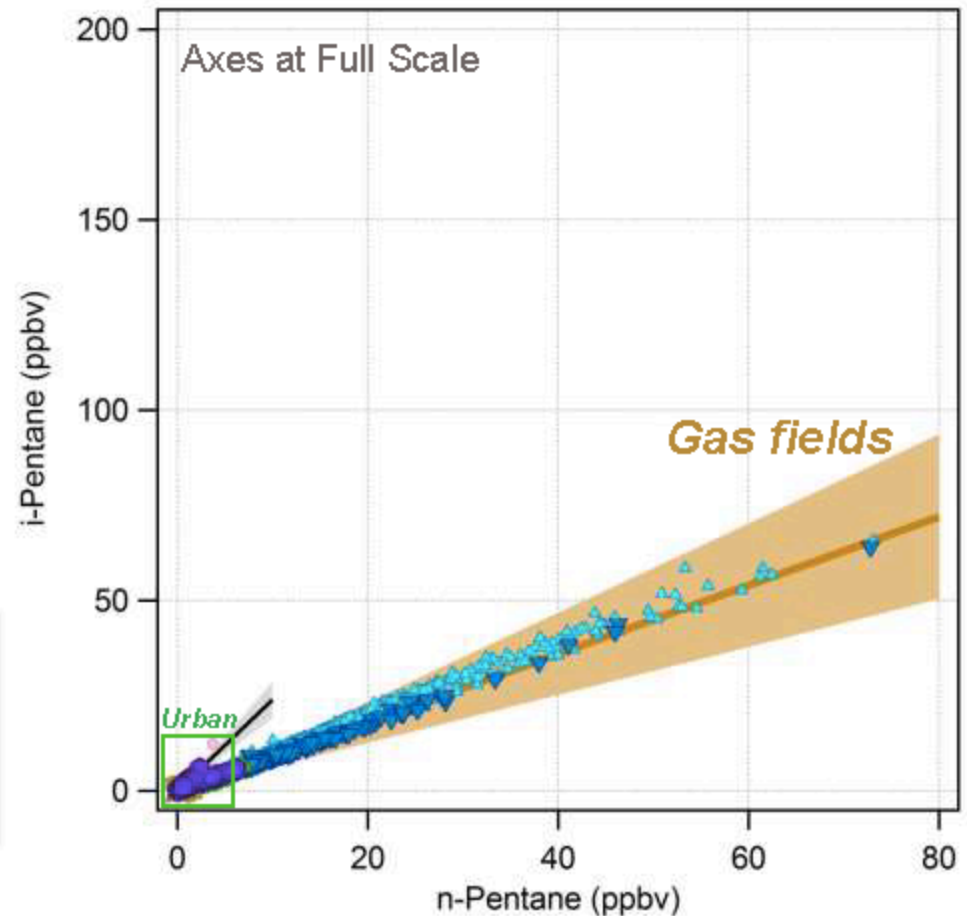
Gasoline combustion vs. Natural gas production

Enhancement Ratios (ER)

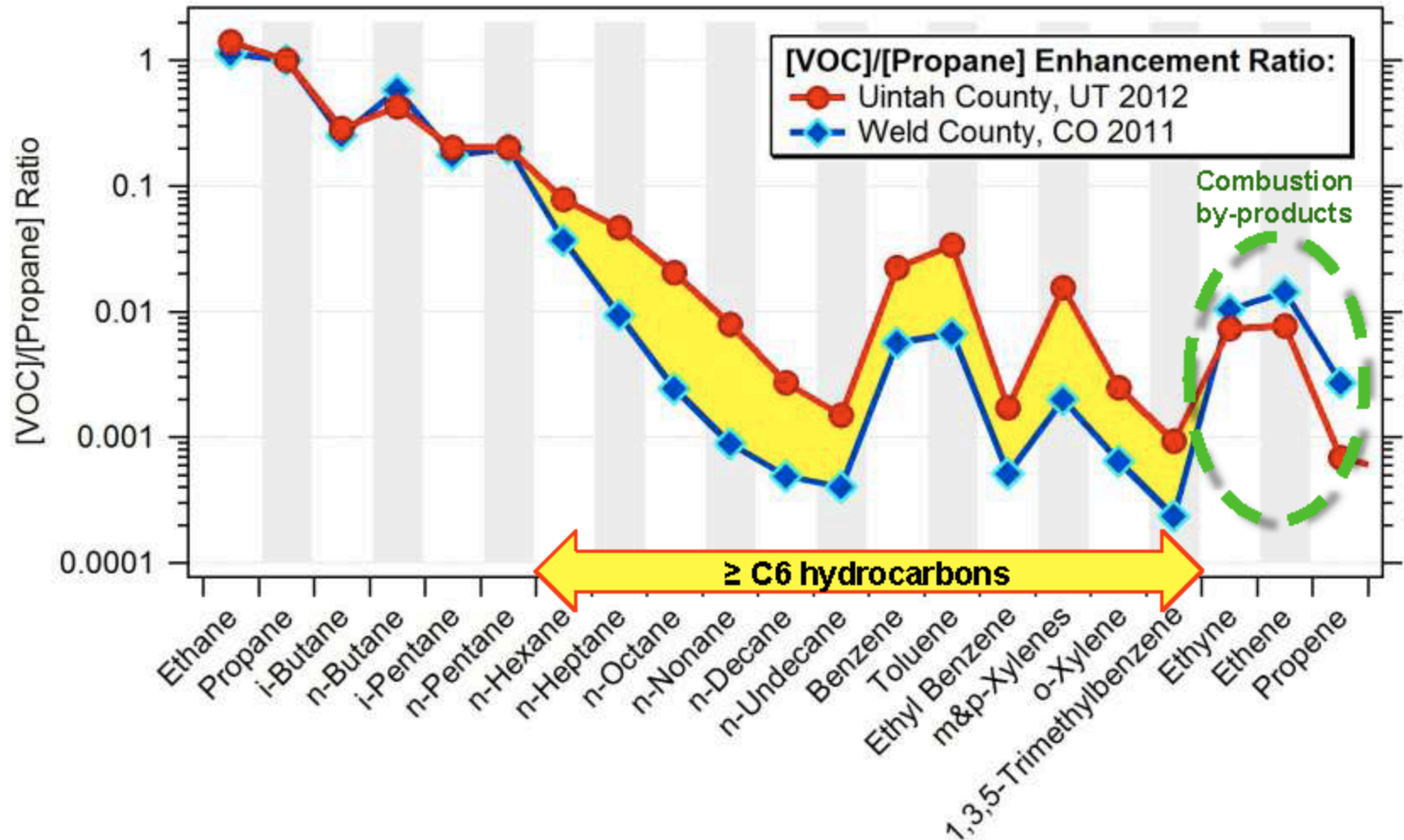
i-Pentane vs. n-Pentane:

- **Urban ER = $2.4 \pm 20\%$**
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- *25 US Cities (Baker, 2008)*
- **Natural Gas ER = $0.88 \pm 20\%$**
- ▲ Uintah Co., UT 2012
- ▼ Weld Co., CO 2011
- *Raw Natural Gas (COGCC rep.)*
- ◆ Fort Collins, CO 2011
- Boulder, CO 2010

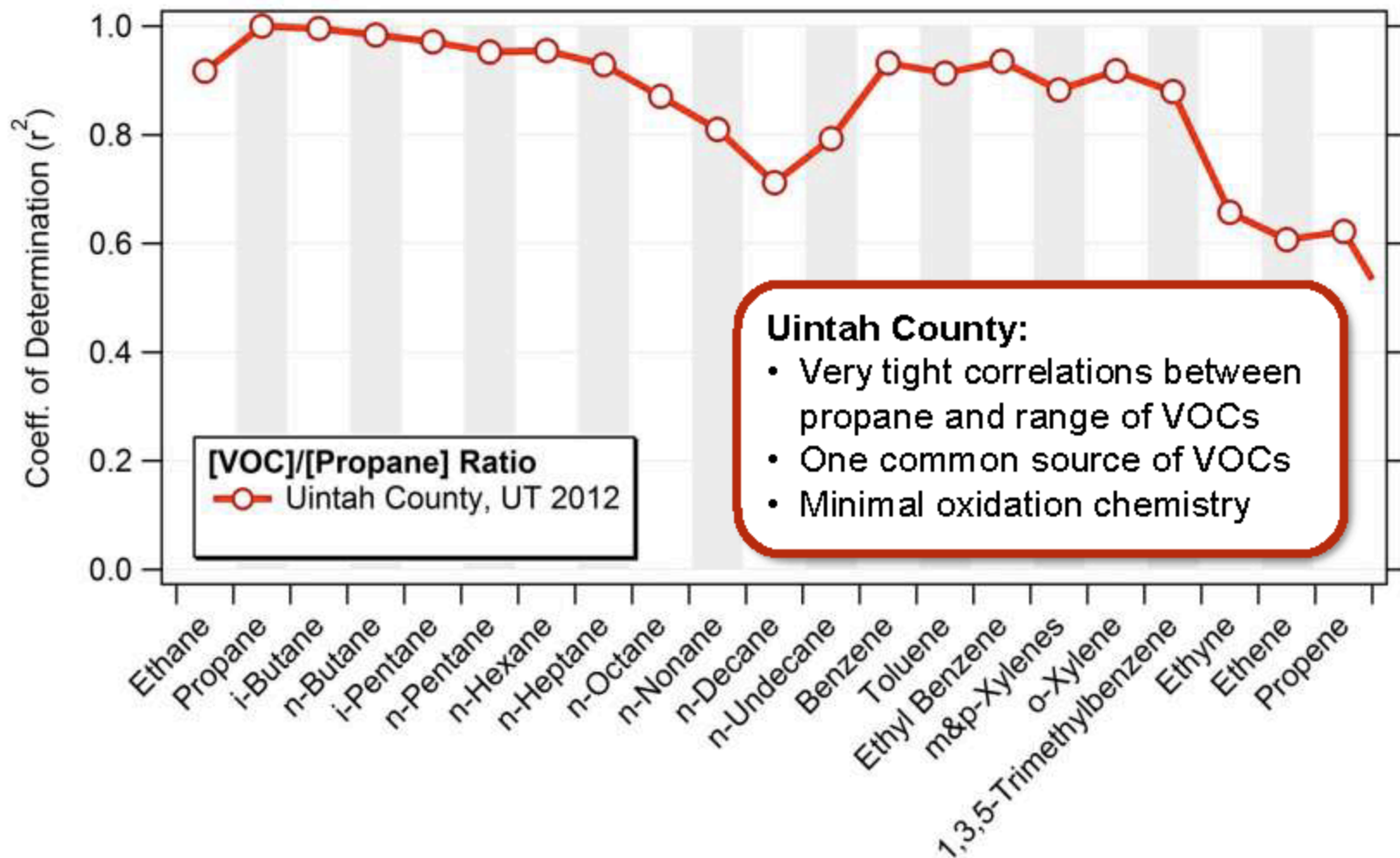
Very large enhancements of alkanes near the areas of oil and gas operations



Emission Source Comparison Colorado vs. Utah Composition

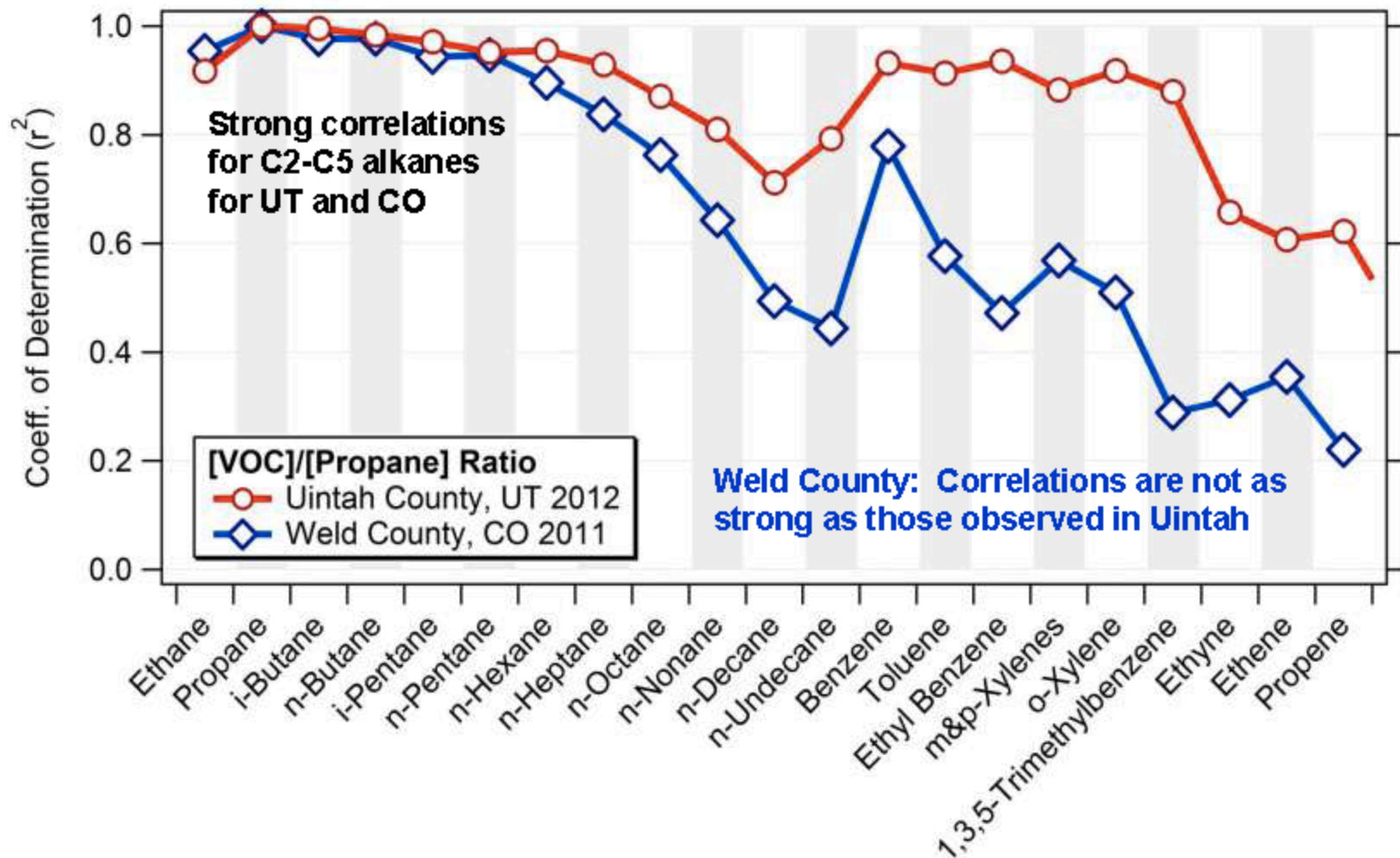


Emission Source Comparison Colorado vs. Utah Source Variability

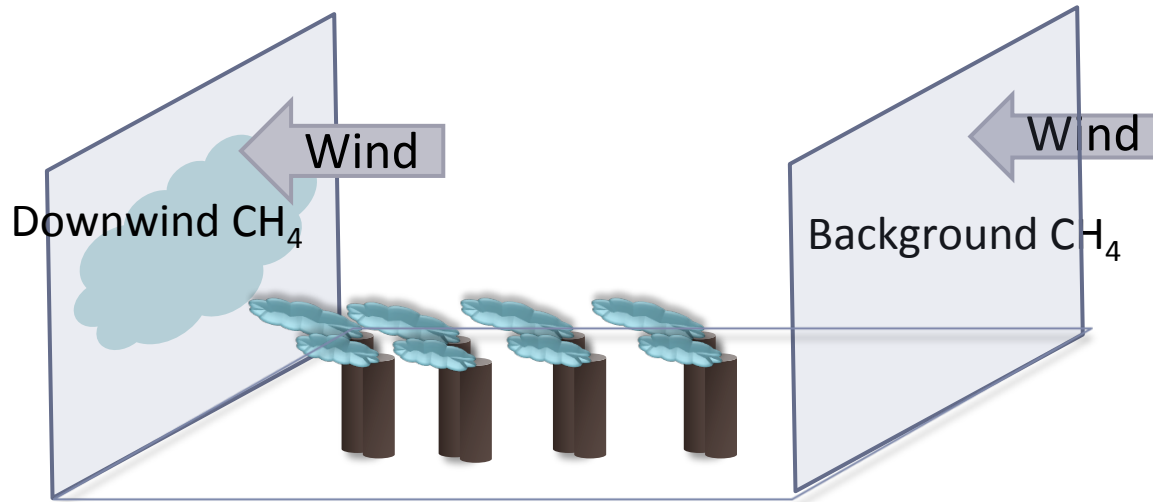


Emission Source Comparison

Colorado vs. Utah Source Variability



Integral Equation of Mass Conservation



mass of CH₄ out of box

mass of CH₄ into box

$$m_{CH_4} = \underbrace{\iint_{CS} \rho_{CH_4} V_n dA_{out}}_{\text{mass of CH}_4 \text{ out of box}} - \underbrace{\iint_{CS} \rho_{CH_4} V_n dA_{in}}_{\text{mass of CH}_4 \text{ into box}}$$

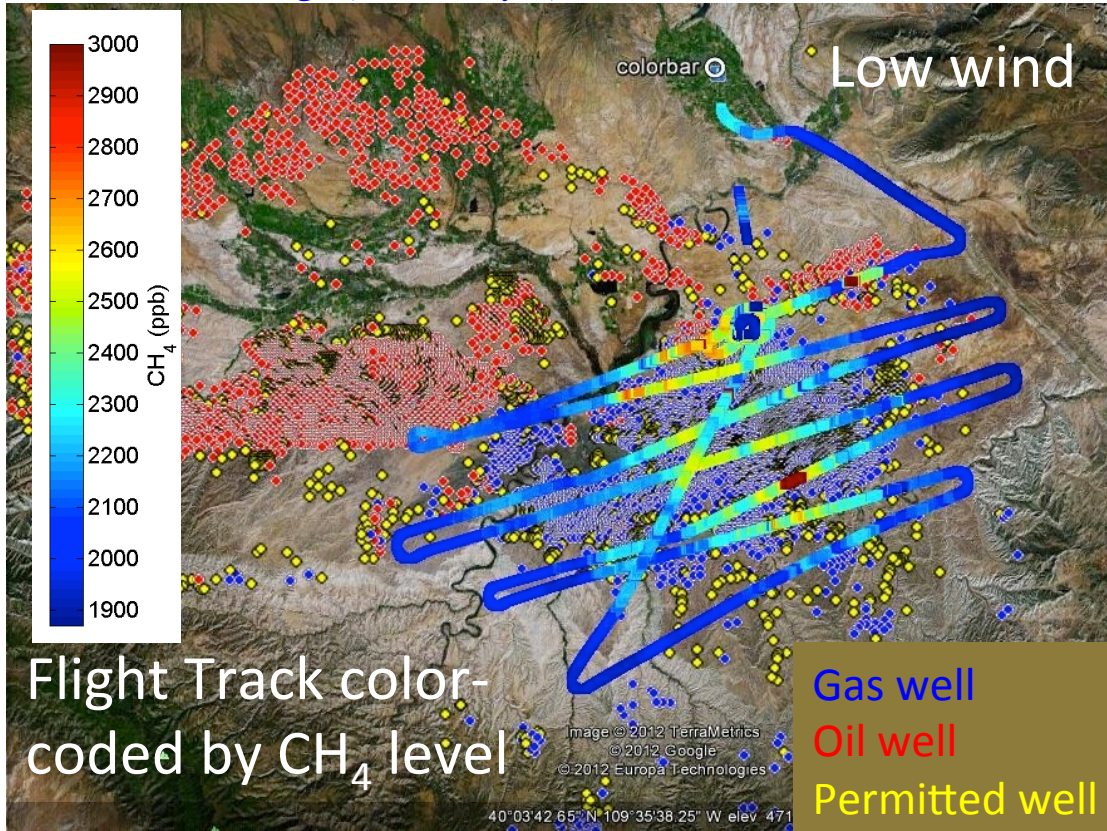
Density of methane (CH₄)

Wind direction normal to plane

Uinta Basin - February 2012

13 flights with in-situ measurements and flask sampling

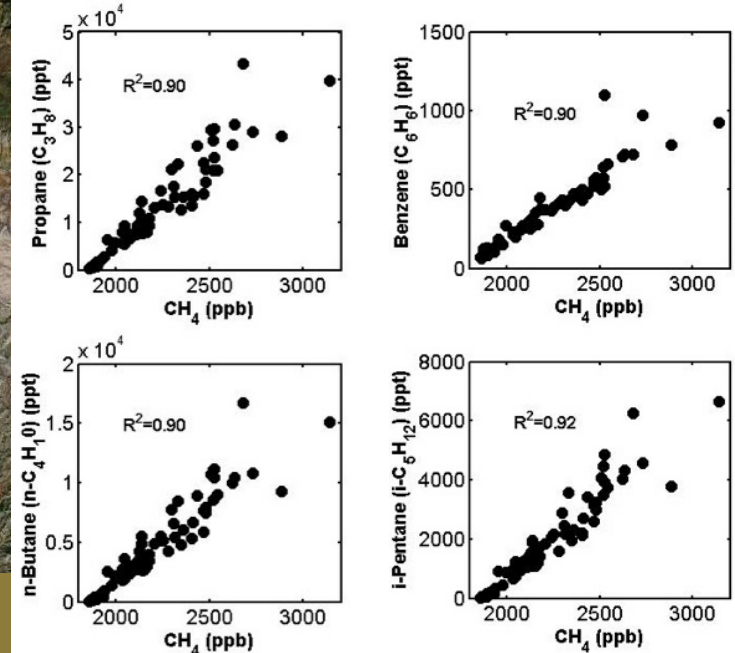
Uintah Basin Flight, February 7, 2012



Oil and Gas production is the main activity in the Uintah Basin.

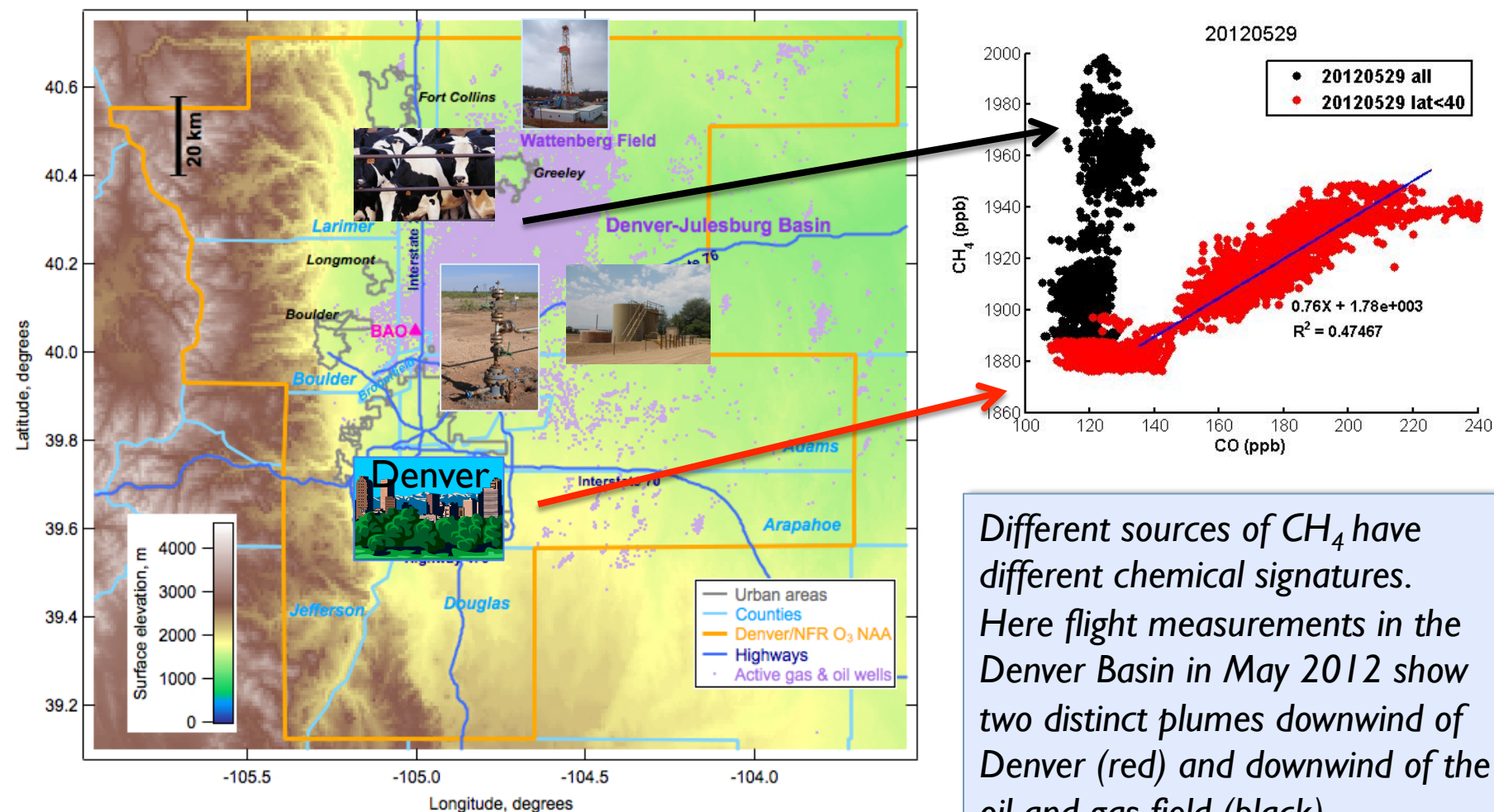
*Multi-laboratory campaign coordinated by
EPA region 8 and State of Utah*

Uintah Basin - February 2012
Aircraft discrete samples data



Strong correlation between methane, the light alkanes and benzene in samples collected in the Uintah Basin in 2012.

Can we use multiple species measurements to do source attribution?

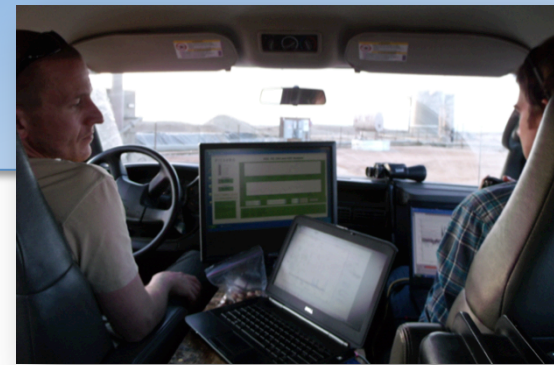


Different sources of CH₄ have different chemical signatures. Here flight measurements in the Denver Basin in May 2012 show two distinct plumes downwind of Denver (red) and downwind of the oil and gas field (black).

Denver Basin, home to > 20,000 oil and gas wells.

▶ A **multi-species** approach is needed to determine the significance of different CH₄ sources as well as to separate different emission processes within the NG industry.

Mobile Lab, Uintah 2012



Displays of measurements

- In situ CH_4 : natural gas marker
- In situ CO_2 , CO : combustion markers
- In situ NO/NO_2 : combustion markers, ozone precursors
- In situ VOCs (aromatics, oxygenates): ozone precursors
- In situ O_3 (two instruments)
- 240 discrete air samples: CH_4 , CO , CO_2 , C_2 - C_8 alkanes, aromatics



NO , NO_2

flasks

VOCs

CH_4 , CO ,
 CO_2

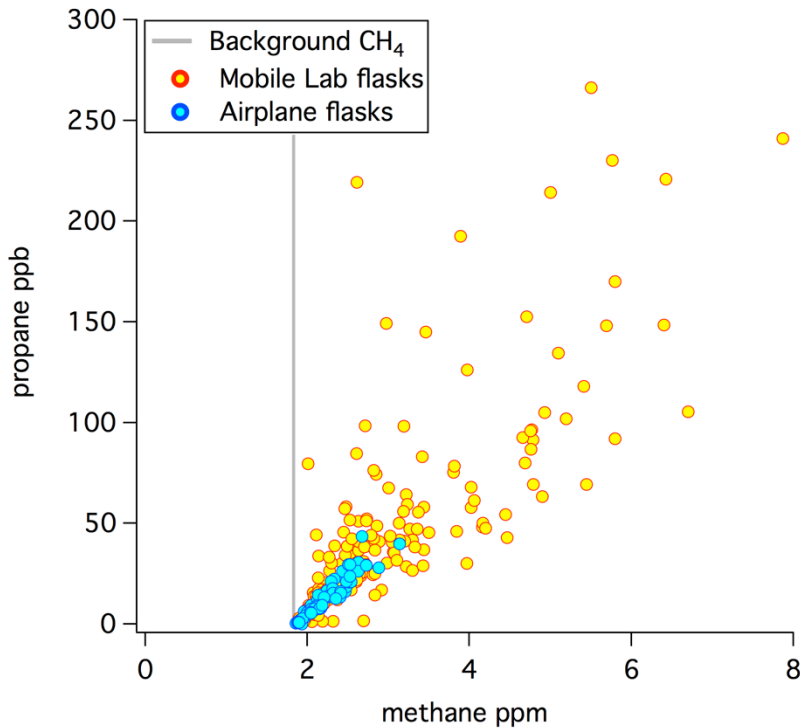


Source identification and characterization

- Tracking chemical signatures of different sources
 - VOCs speciation (Alkanes, Aromatics, Alkenes)
 - VOCs/CH₄ ratios: fugitive emissions versus condensate/oil vapors
 - NO_x/CO₂ ratios: emission factors for various point sources in the Basin
 - CO/CO₂, (CO+CO₂)/(CO+CO₂+CH₄): combustion efficiencies
- Comparison with WRAP upstream O&G emissions inventory
 - Using measured emission factors and activity data
 - Using airplane and NOAA HRDL met fields measurements

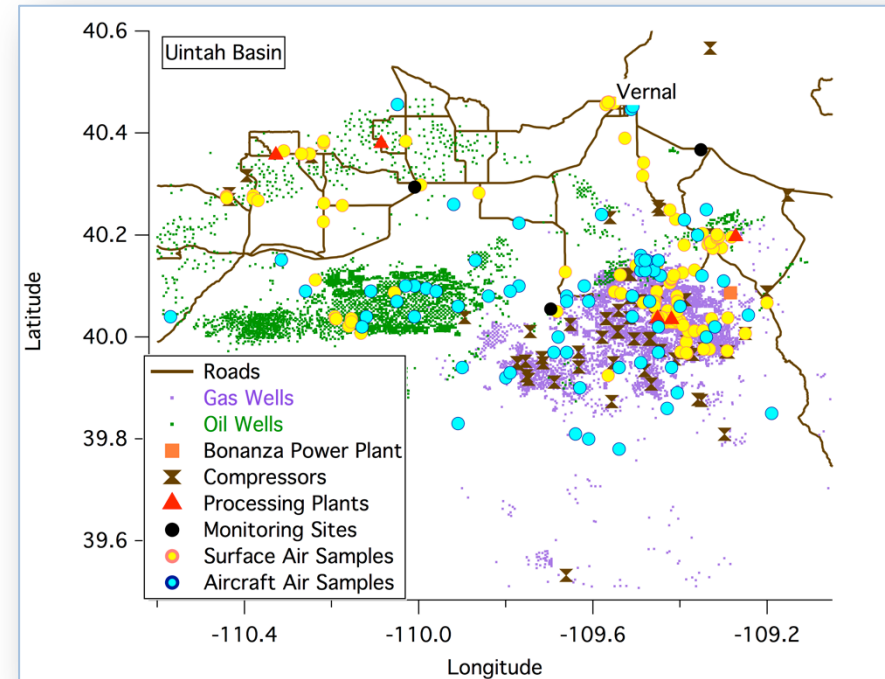
Note: Similar studies carried out in DJB over past several years

Discrete Air Samples: CH₄, CO, CO₂ and VOCs

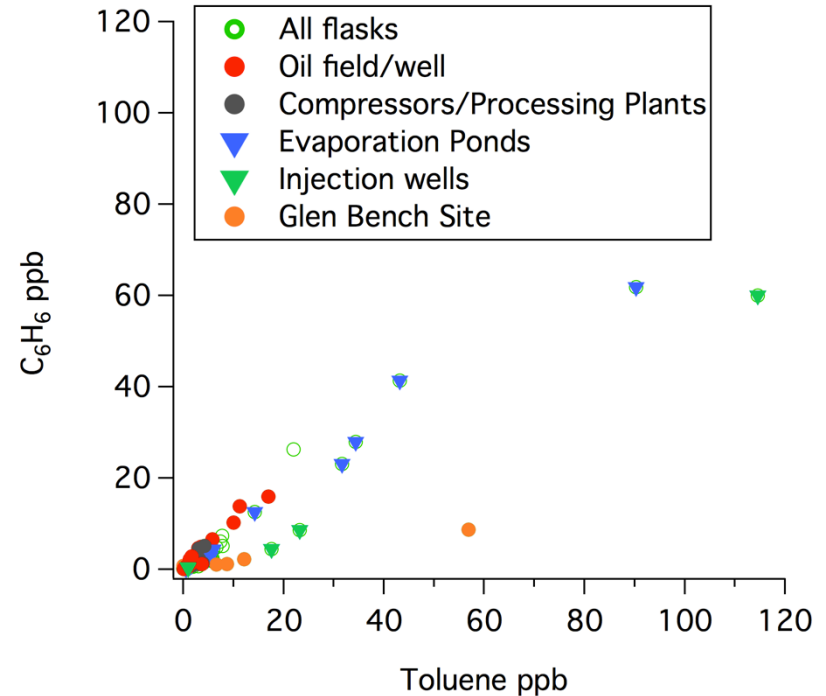
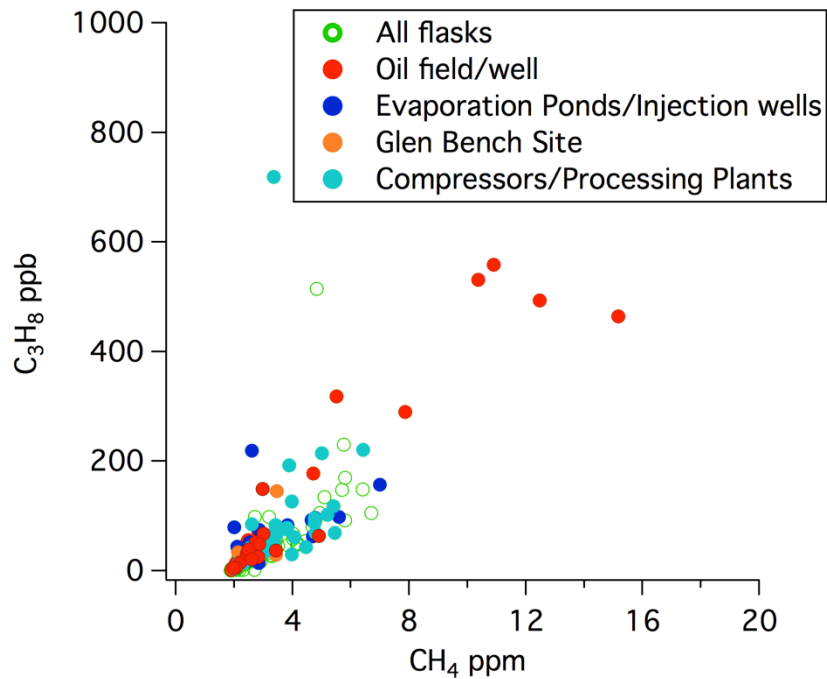


Example of flasks data:
The airplane sampled the mixture of emissions while the Mobile Lab was able to target specific sources.

- Mobile Lab: 240 flasks
- Airplane: 81 flasks
- Collected in various areas in the oil and gas Basin



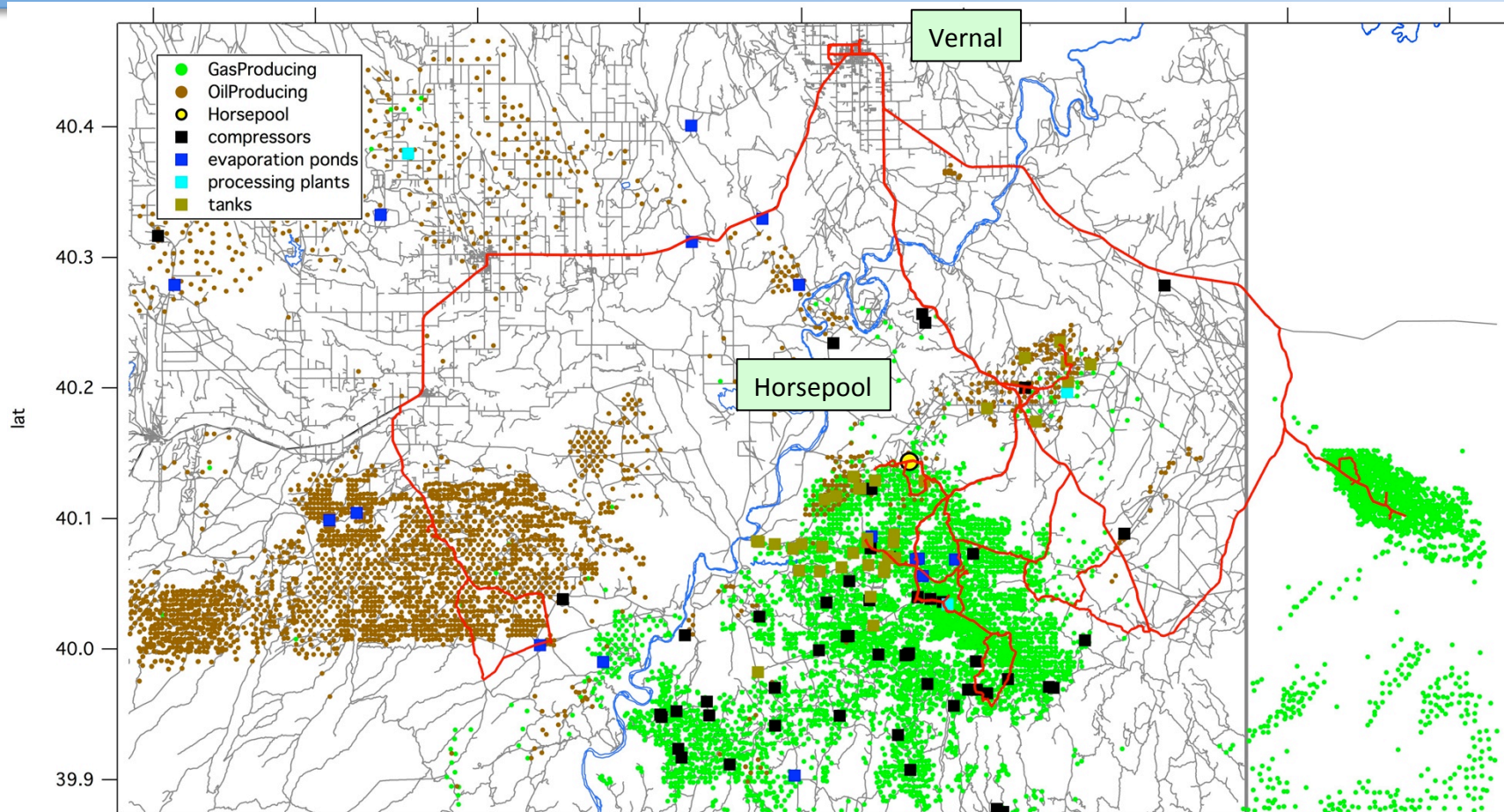
Mobile Lab flasks data shows a mixture of VOCs



Targeted sampling of regional enhancements and specific point sources plumes show that alkanes are the most enhanced VOCs.

Aromatics levels are also elevated (> 2 ppb) downwind of certain sources.

Measured levels are dependent on proximity to the center of the plume and dispersion conditions when the sample was collected.



- Mobile Laboratory sampled wide variety of individual sources
- Source composition useful for comparison with data from Horsepool and aircraft: which sources are the largest contributors?

Oil evaporation:
Aromatics,
cycloalkanes

Gas evaporation
methane

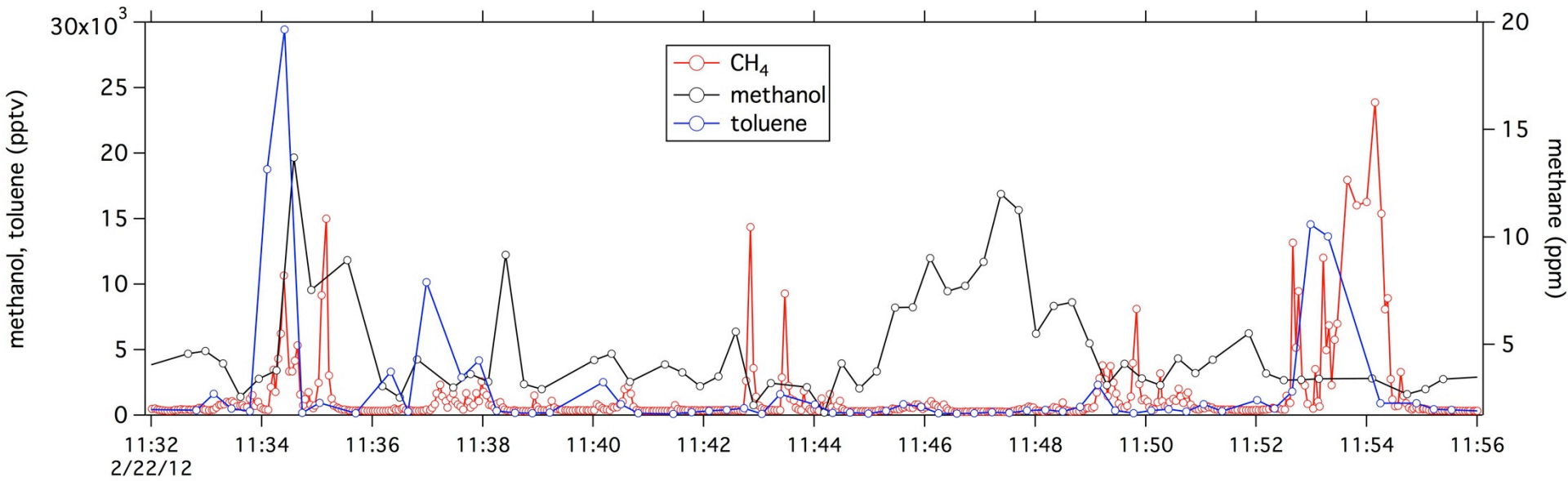
Leaks methane
methanol

Condensate
tanks

wellhead

Methanol tank

separator



Fast-response VOC measurements allow separate measurements of individual sources

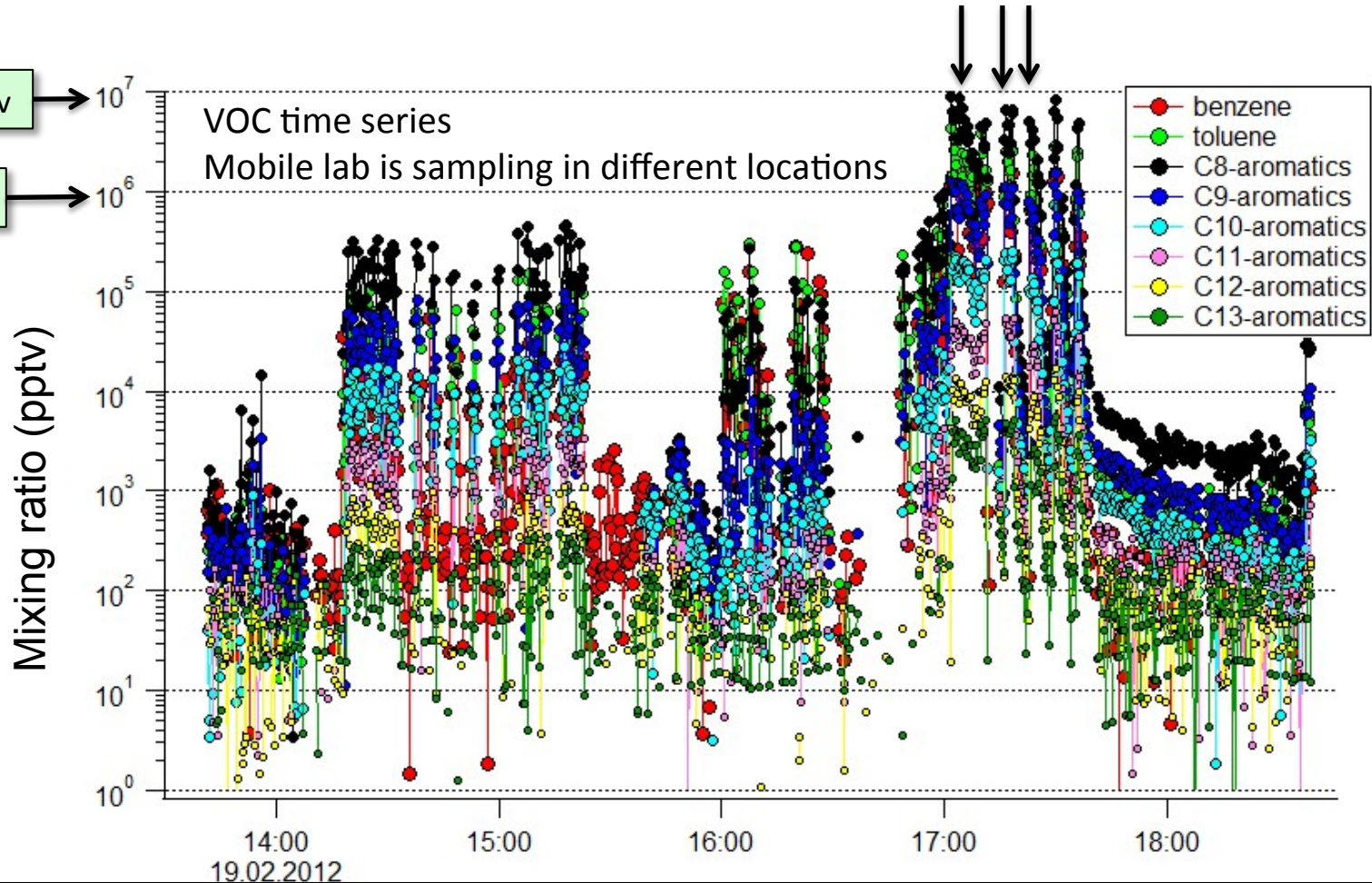


Measurements at new well near Glen Bench:

1. Well was fracked on February 8-9
2. Flow-back pond was used
3. Mobile laboratory went by 11 times between Feb 14-28; well not completed on Feb 28

Results from one day (Feb 19):

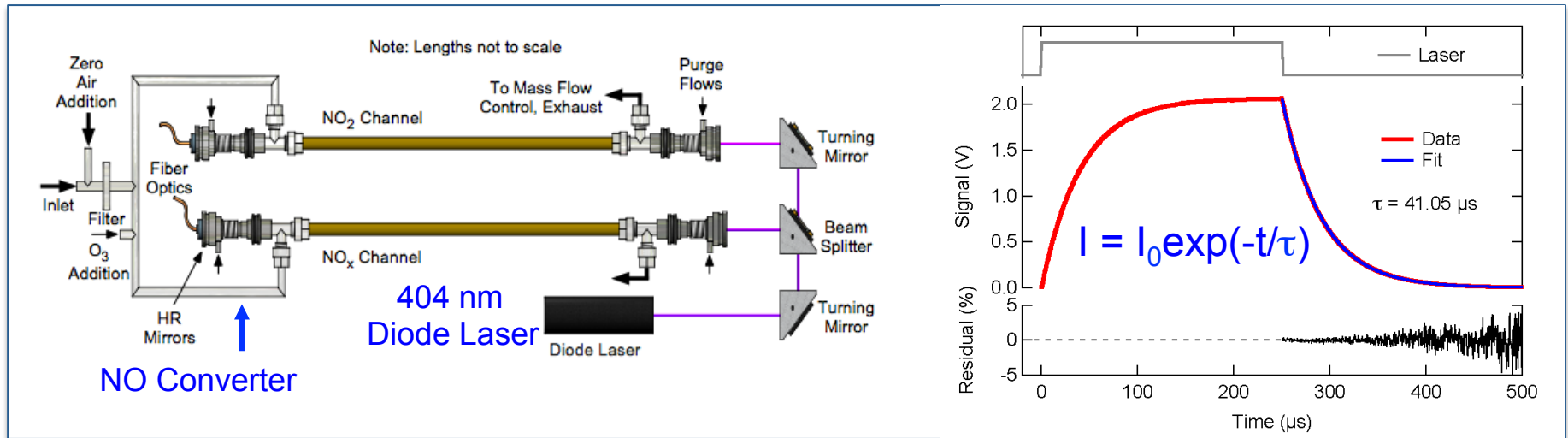
- Repeated drive-by of the new well site
- Aromatics concentrations peak to 1-10 ppmv **every time** the mobile lab is downwind from flow-back pond



High concentration of aromatics (1-10 ppmv) near flow-back pond

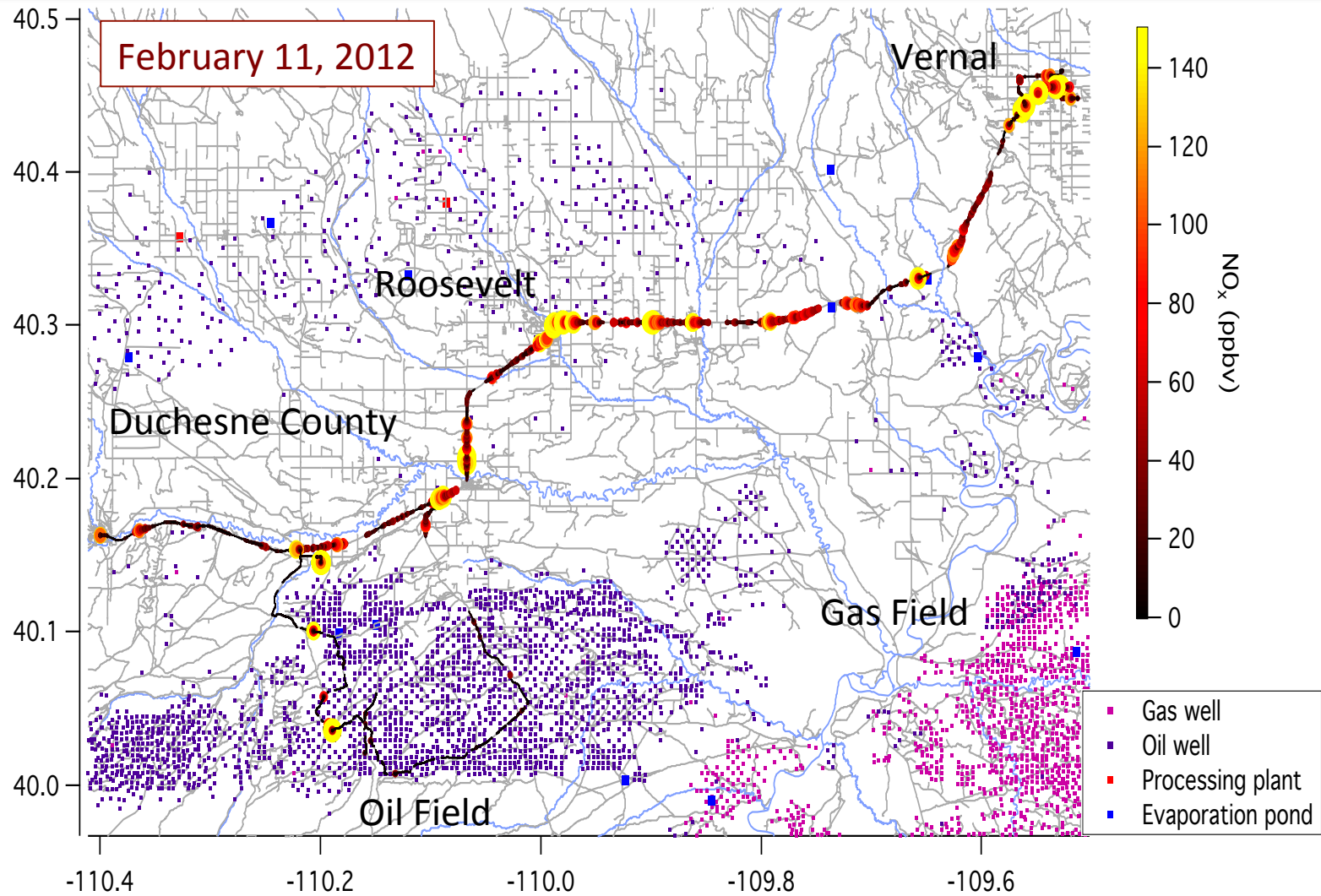
Cavity Ring-Down NO_2 , NO_x (O_3) Measurements

Peter Edwards, Steven Brown, William Dubé



- Measure NO_2 directly by 404 nm optical absorption
Effective path length = 12 km; Sensitivity to NO_2 = 0.05 ppb, 1 s; Accuracy = 5%
Measured signal in Utah: 0.5 – 10 ppb background, 10-1000 ppb in plumes
- Measure NO_x (=NO+ NO_2) by quantitative (98%) conversion of NO to NO_2 in excess O_3
 $\text{NO} + \text{O}_3 \rightarrow \text{NO}_2 + \text{O}_2$
Simple and accurate scheme for measurement of total NO_x
- For some drives, measure O_3 , rather than NO_x , by addition of excess NO to second channel
- Instrument weight, power consumption, size appropriate for deployment in Mobile Lab

Example Mobile lab NO_2 , NO_x and O_3 data for single drive

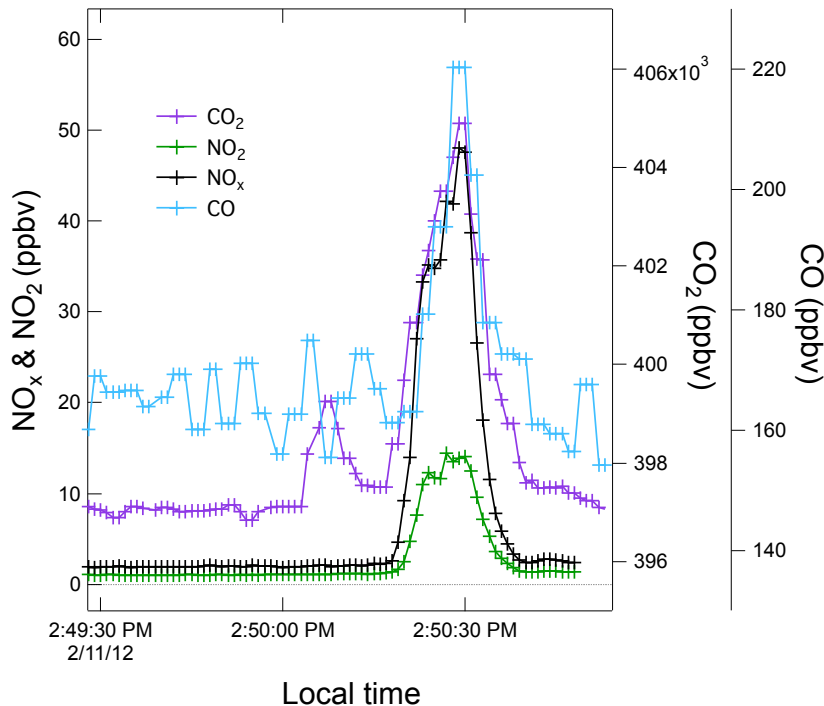


10 drives measuring NO_2 and NO_x

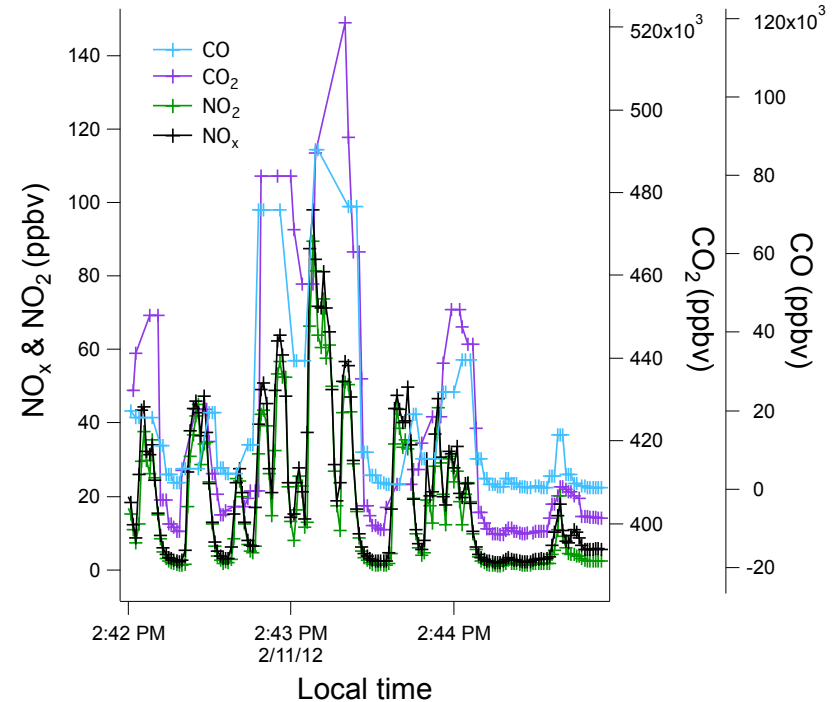
9 drives measuring NO_2 and O_3

Example NO_x Emissions Factors Calculation

Truck Plume



Pumpjack Plume



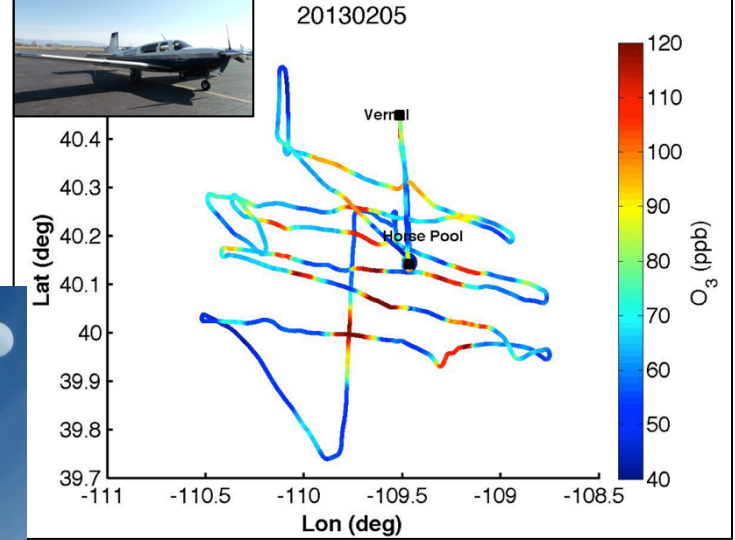
NO_x vs CO₂ useful for NO_x emissions vs. fuel use or number of engines / vehicles
NO_x vs CO useful for comparison to emissions inventories (e.g., CO often used as urban tracer)
NO₂ vs NO_x useful for direct NO₂ emission *if* this ratio is large

NOAA measurements & analysis of chemical and meteorological processes in O&G regions

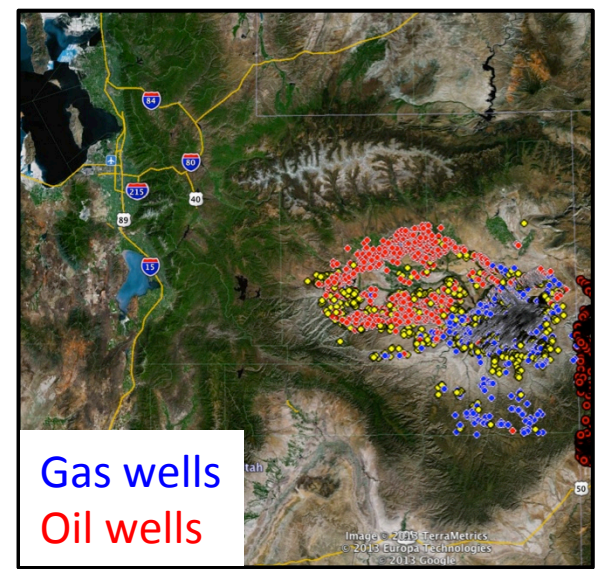
Examples of measurements that probe...

- Ozone distributions
 - Chemical processes leading to ozone formation
 - Basin flows and boundary layer dynamics
1. Tethered sondes + aircraft, Uintah winter 2013 (Schnell et al.)
 2. Surface sampling, Uintah winter 2012/2013 (Roberts et al.)
 3. Surface sampling, DJB winter 2011/ Uintah winter 2012 (Gilman et al., 2013)
 4. Surface remote sensing, Uintah winter 2012/2013 (Senff, Brewer, Banta, et al.)

Wintertime Ozone in NE Utah

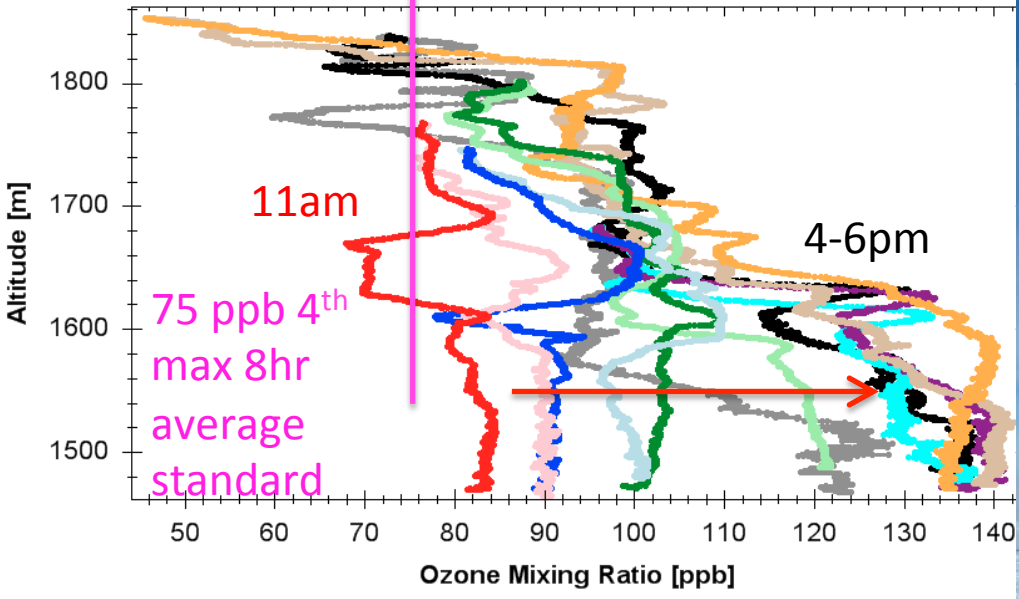


Tethered sonde and airborne O_3 measurements in the Uinta Basin in February 2013



Ozone Mixing Ratio at Fantasy Canyon, 2/5/2013

- Hour 11 Asc
- Hour 11 Des
- Hour 12 Asc
- Hour 12 Des
- Hour 13 Asc
- Hour 13 Des
- Hour 16 Asc
- Hour 16 Des
- Hour 17 Asc
- Hour 17 Des
- Hour 18 Asc
- Hour 18 Des



Under very strong temperature inversion, the Uinta Basin experienced very high ozone levels at the surface (up to 150 ppb). Oil and gas operations are the main activity in the region.

R. Schnell et al.

UBWOS 2012 & 2013: Horse Pool Intensive Site

How is Ozone made in the Lower Atmosphere?



VOCs
NO_x

OH Radicals
Cl Atoms

O₃

*Ingredients
(Sources)*

*Initiators
(Contributing
Conditions)*

*When, Where
(Distribution)*

UBWOS 2012 & 2013: Horse Pool Intensive Site

Radio-
meters

IBBCEAS (NO_2 , HONO)

ARNOLD (NO_3 , N_2O_5 , NO_x , O_3)
Acid CIMS (HONO, HCl, HNO_3 , org acids)

20m

10m

NO_y species
 ClNO_2

VOCs

PMEL

Doppler
LIDAR

Ozone
LIDAR

DOAS

UBWOS 2012 & 2013: Horse Pool Intensive Site

Measurement Goals

- ❑ VOC and NO_x sources and levels
Ingredients
- ❑ Possible radicals sources, CH₂O, HONO, ClNO₂
Initiators
- ❑ Aerosol/Snow Chemistry
Contributing Conditions
- ❑ 3-D distribution of O₃
When, Where
- ❑ Transport within, and stability of, the boundary layer
Contributing Conditions

2012: No snow → Lower O₃

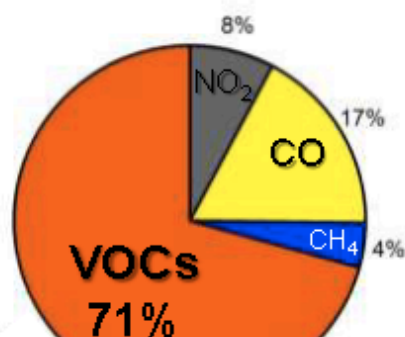
- P. M. Edwards et al., ACPD, 2013

2013: Snow → Elevated O₃

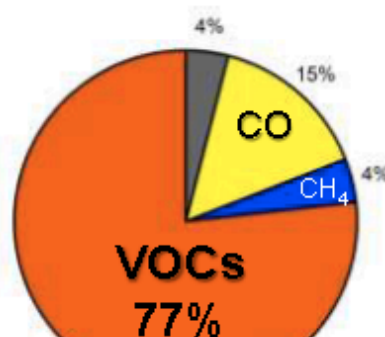
OH reactivity: Comparison of composition

Total OH reactivity

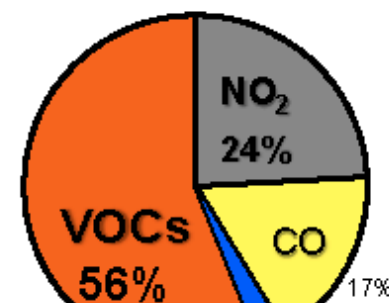
Uintah County, Utah



Weld County, Colorado

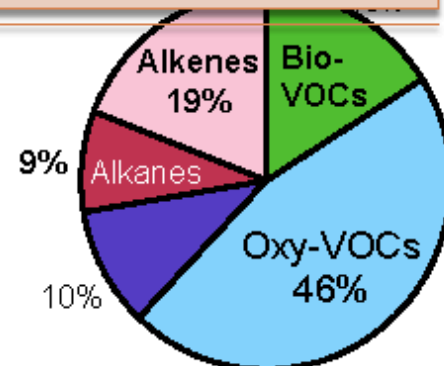
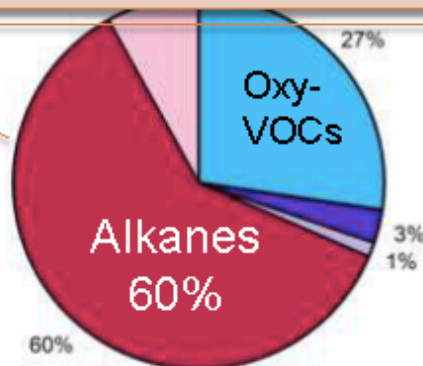
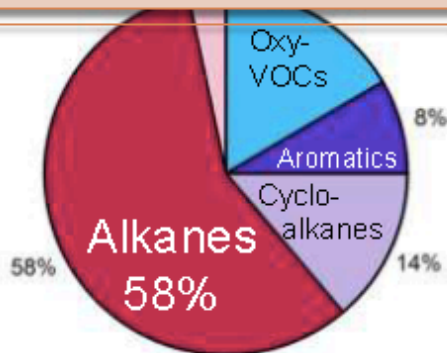


Pasadena, California
Spring - Summer



Based on "propane term" of multivariate fit analysis for Weld County, Colorado:
Oil and natural gas operations contribute 57% of VOC OH reactivity

OH reactivity of VOCs



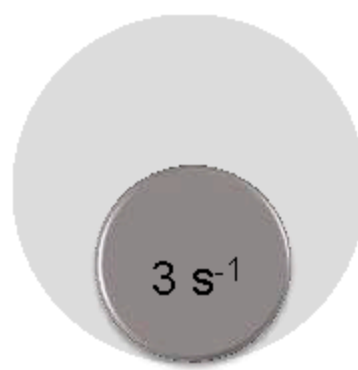
OH reactivity: Comparison of median values

Total OH reactivity

Uintah County, Utah



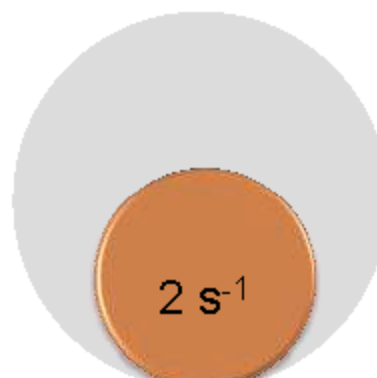
Weld County, Colorado



Pasadena, California
Spring - Summer



OH reactivity of VOCs

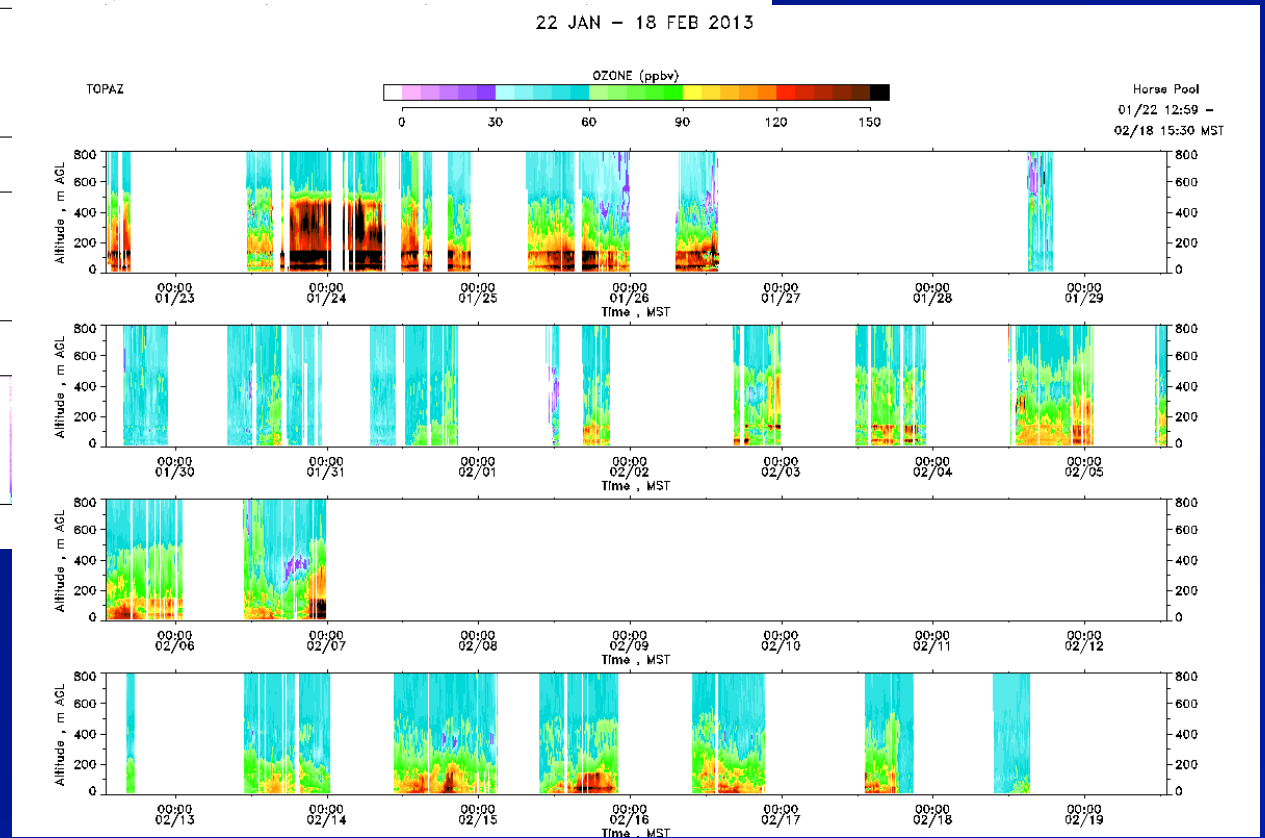
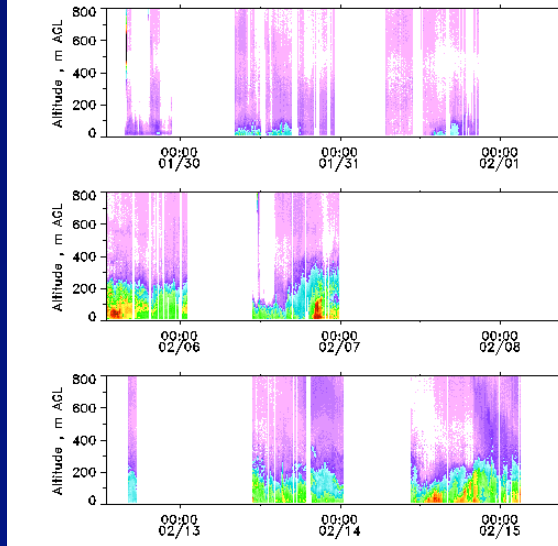
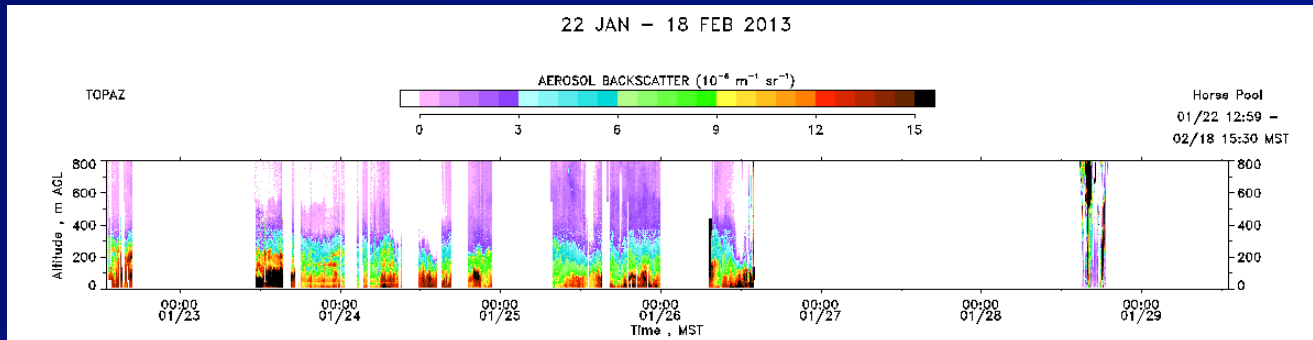


Areas scaled by median OH reactivities

Dynamics studies: Uintah 2013

- Depth of pollution layer
- Spatial variability of source activity
- Mixing mechanisms
- Patterns of horizontal flow evolution
- Horizontal spread of pollutants

Time-height O₃, aerosol – Uintah 2013

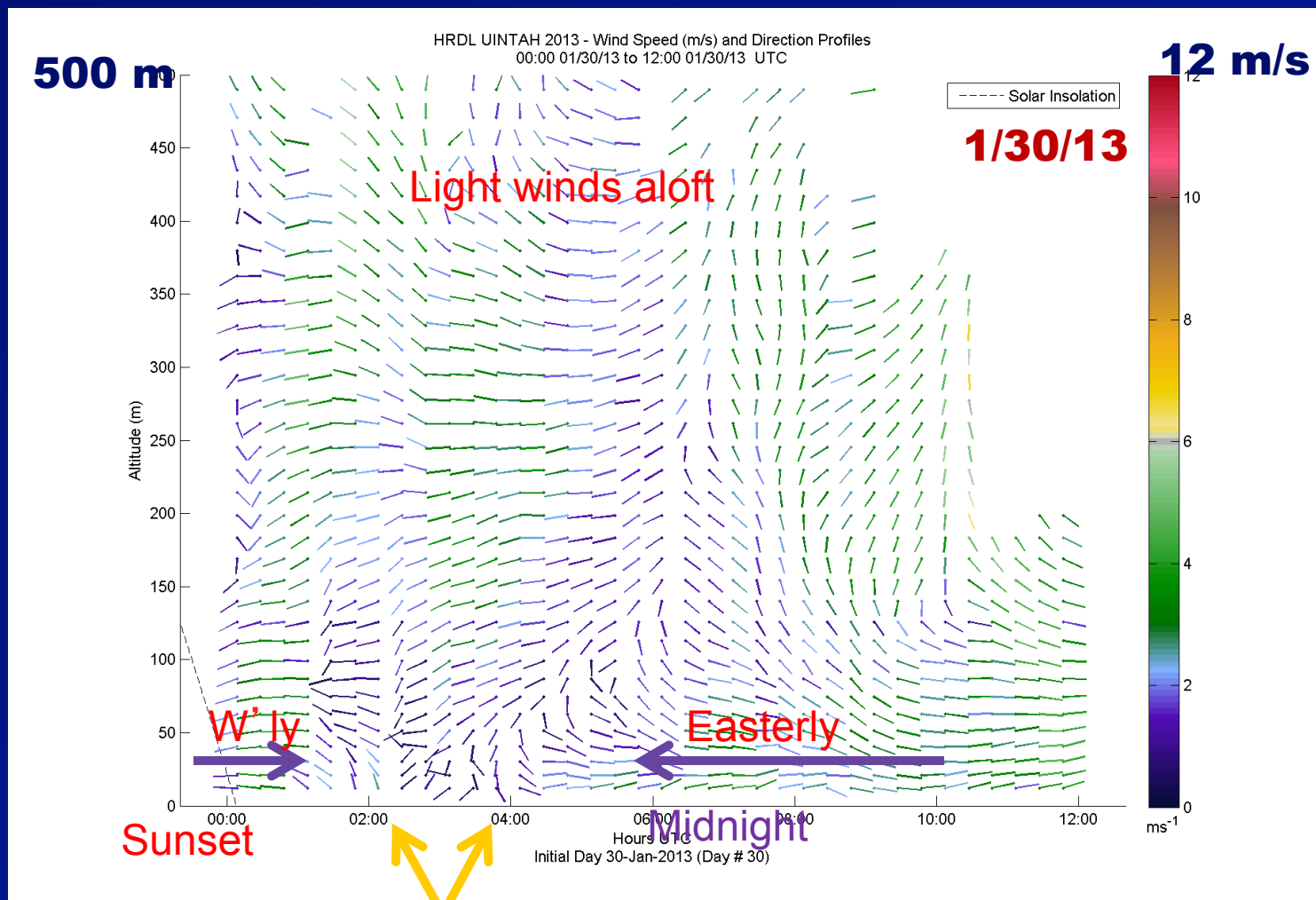


Depth of polluted layer – deepens thru event

C. Senff et al.

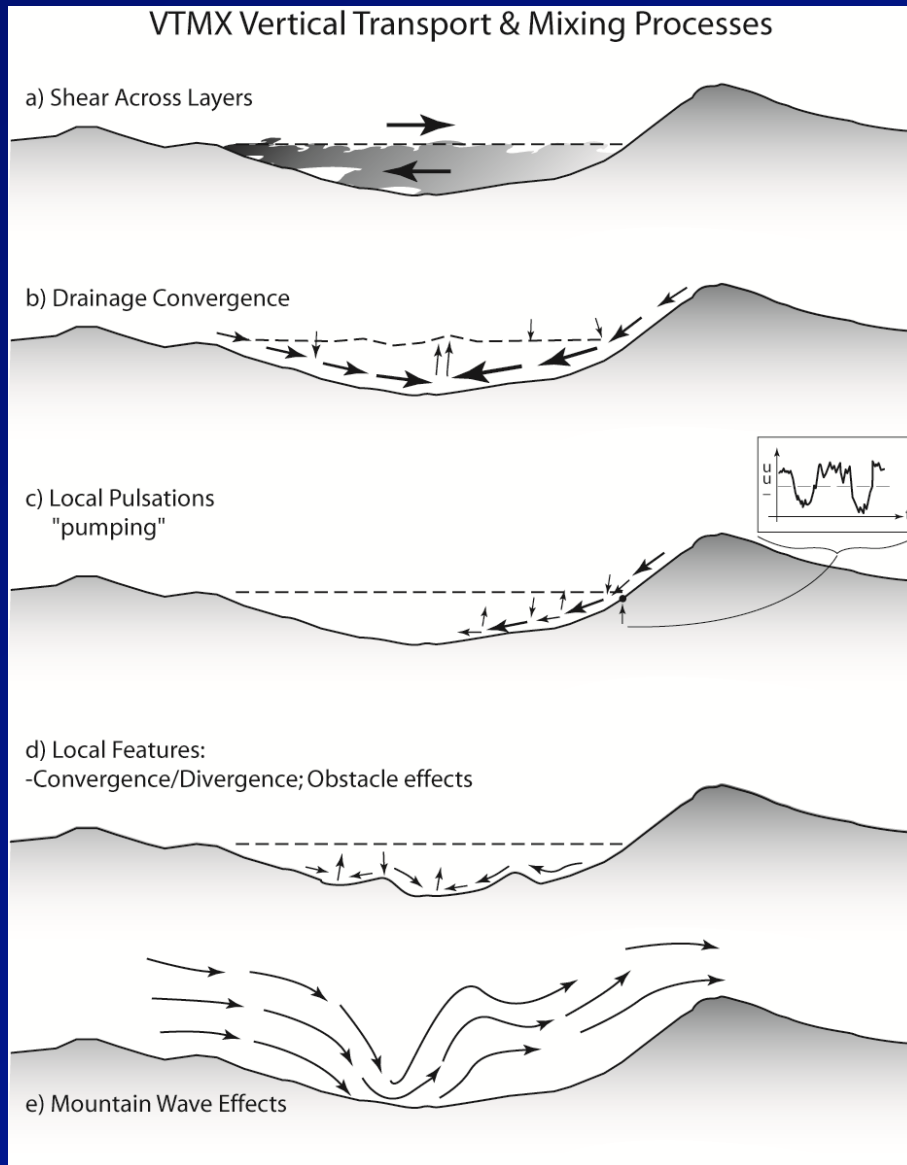
HRDL profiles – nighttime, Uintah 2013

Time vs. height – wind speed color coded



COLD SURFACE

How can we get O₃ (e.g.) up to 300 - 400 m when stable?



Vertical mixing processes in a basin cold pool: complex, for a cold surface

Banta, R.M., et al., 2004: Nocturnal low-level jet in a mountain basin complex. Part I: Evolution and implications to other flow features. *J. Appl. Meteor.*, 43, 1348-1365.

NOAA modeling to understand emissions and impacts on air quality in O&G regions

Use meteorological/chemical models in forward and inverse approaches to test understanding of emissions, chemical processes, and atmospheric dynamics

1. Regional chemical-transport model, California/Utah/Colorado (Ahmadow and McKeen)
2. Inverse model, California (Brioude)

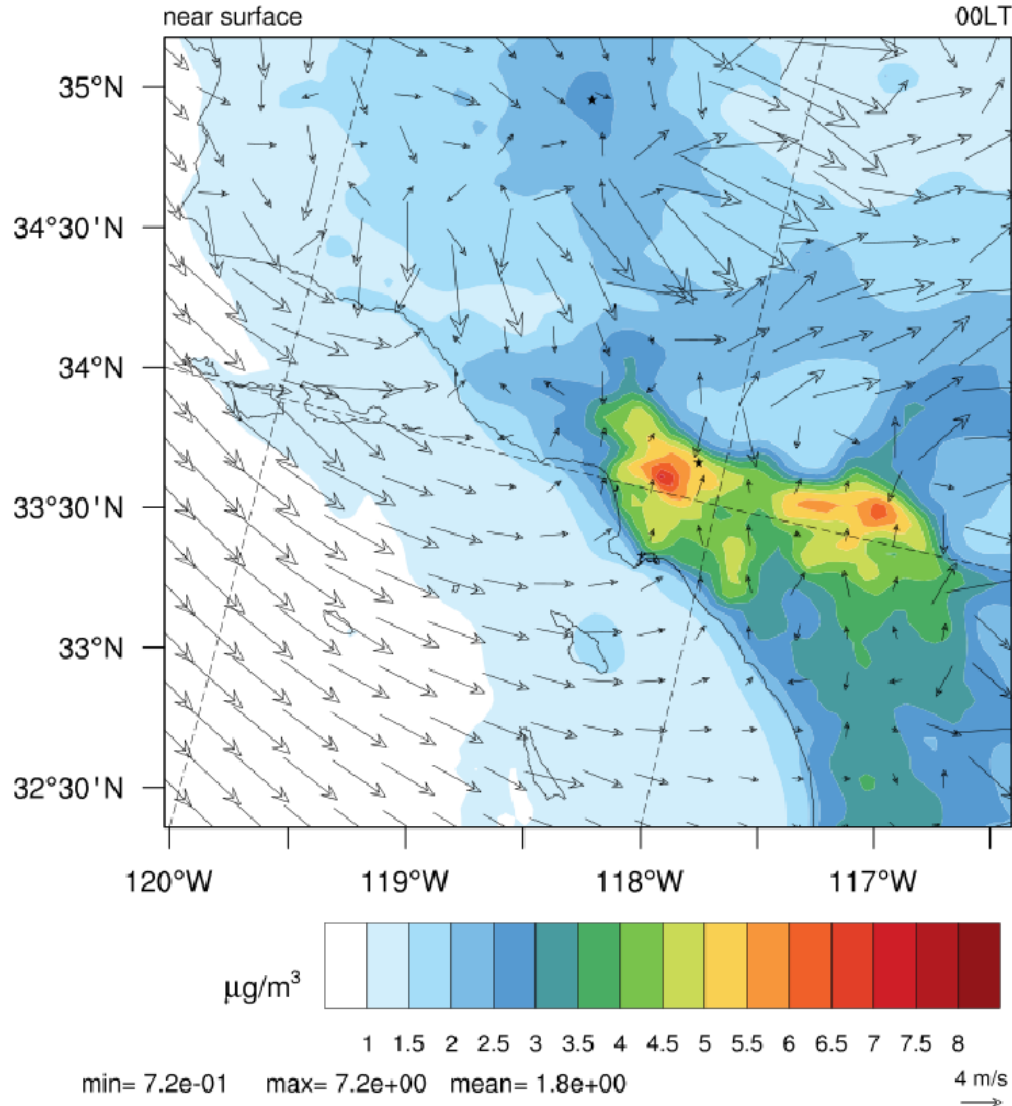
WRF-Chem Modeling

Stuart McKeen and Ravan Ahmadov

- Coupled meteorology-chemistry modeling for CalNex-2010, UBWOS-2012 and 2013 studies
- Secondary organic aerosol simulations by using volatility basis set approach in WRF-CHEM
- Evaluation of different anthropogenic emission inventories
- Extensive evaluations against aircraft and surface measurements

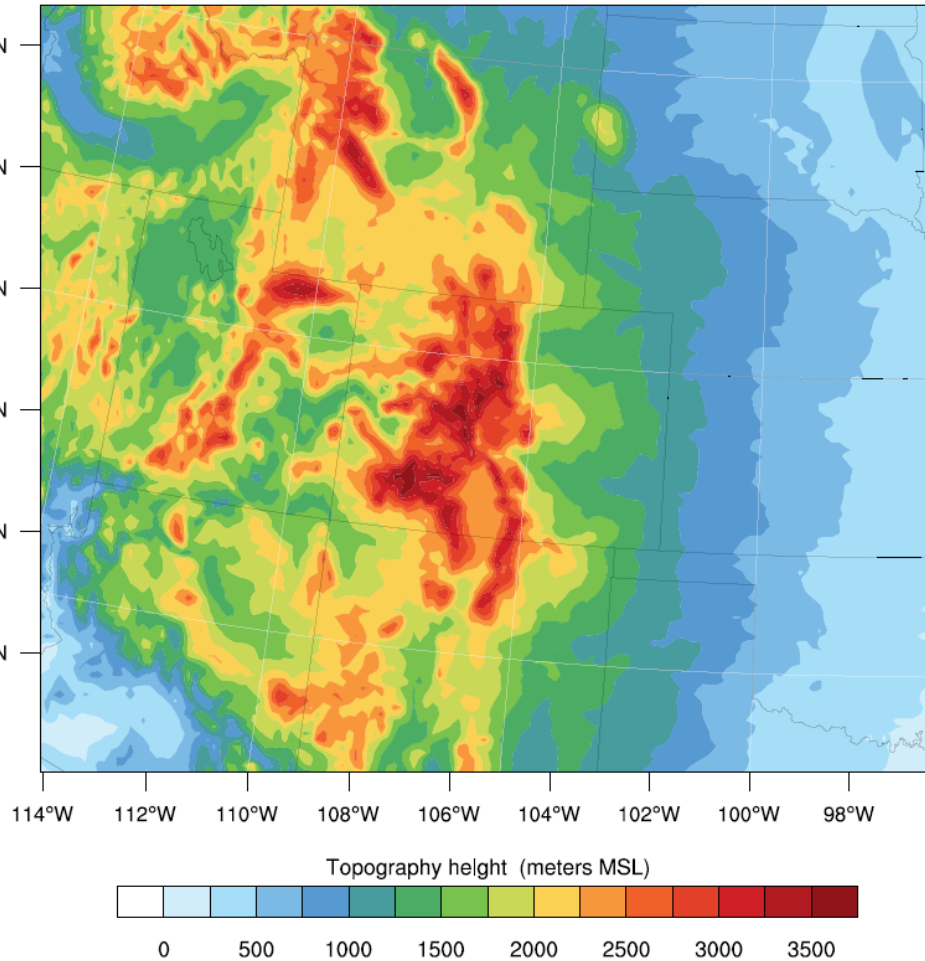
Organic aerosol simulations for LA Basin

Averaged fields 05/14-06/15, 2010

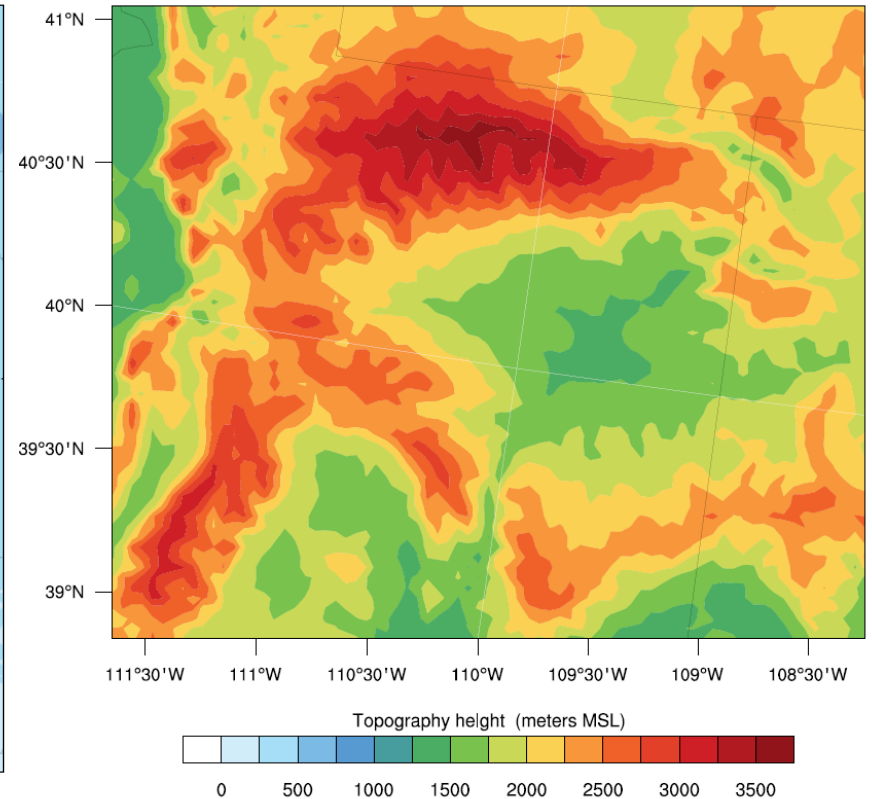


Nested WRF-CHEM domains for the UBWOS modeling study

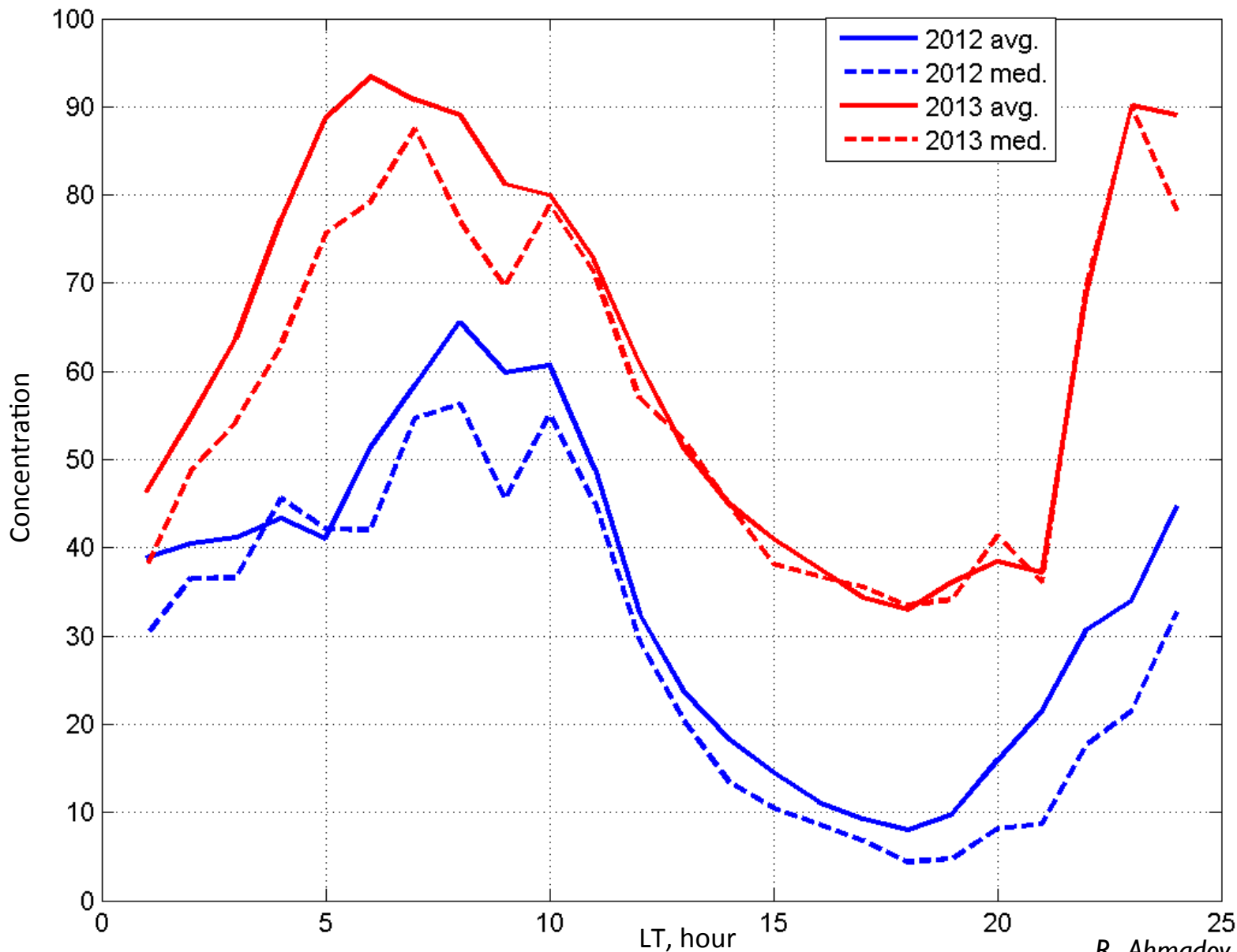
12km resolution



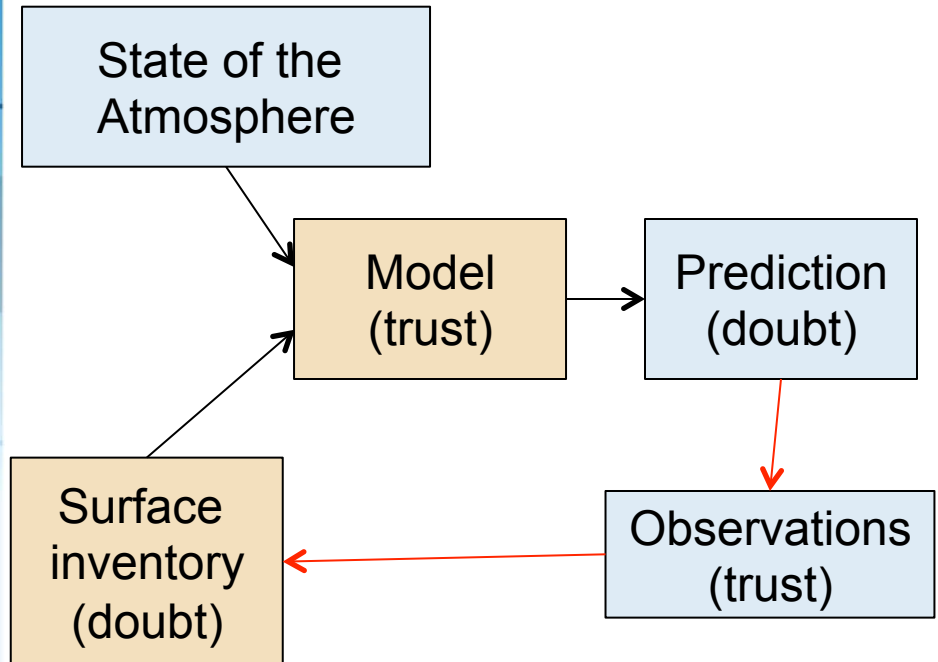
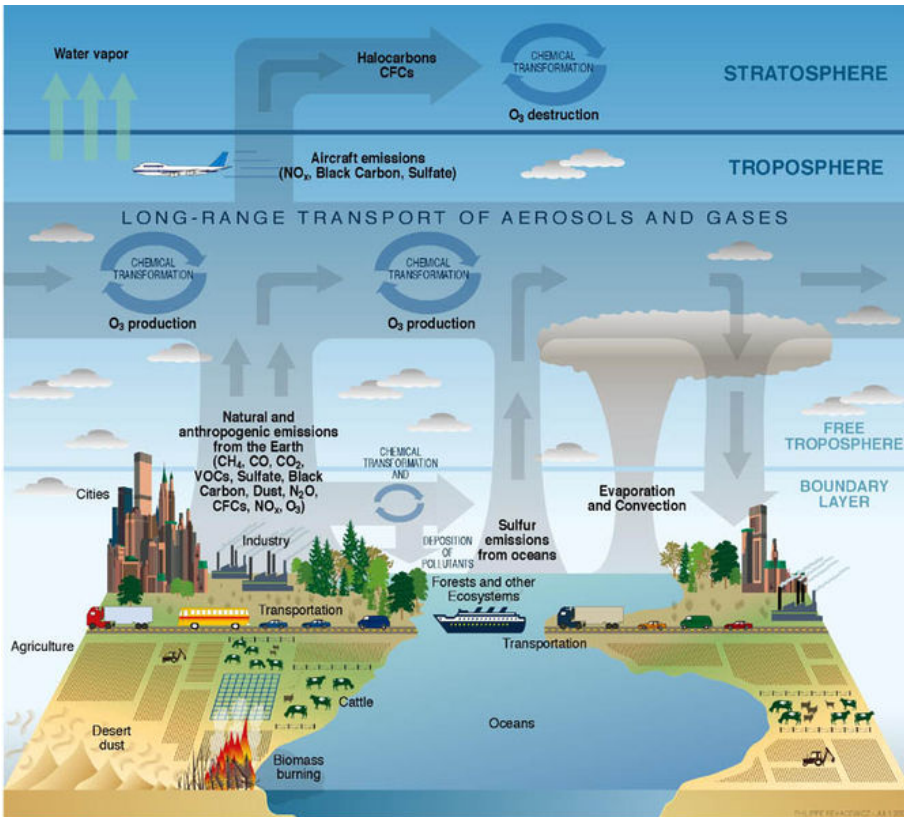
4km resolution



Diurnal cycle of a modeled passive tracer at the HorsePool site, February, 2012, and 2013

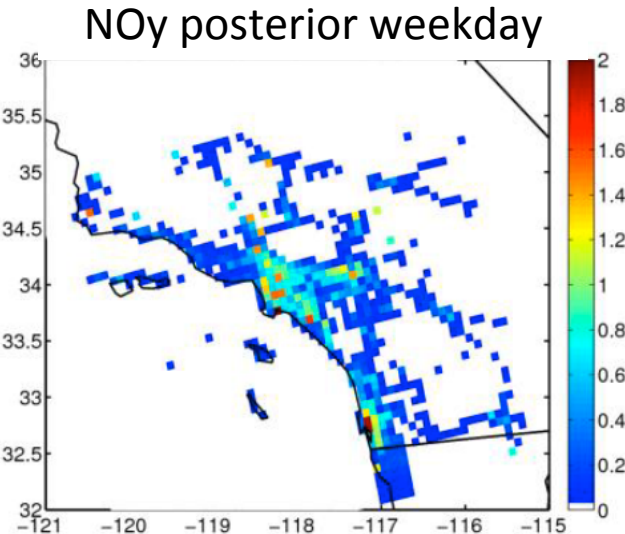
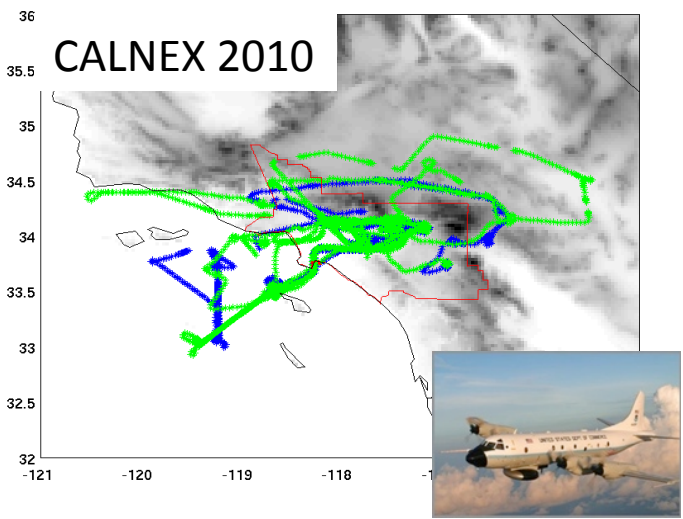


Mesoscale inverse modeling to evaluate emissions



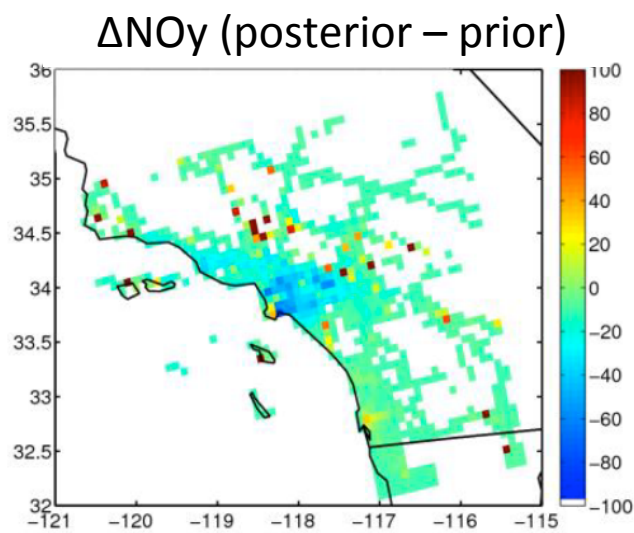
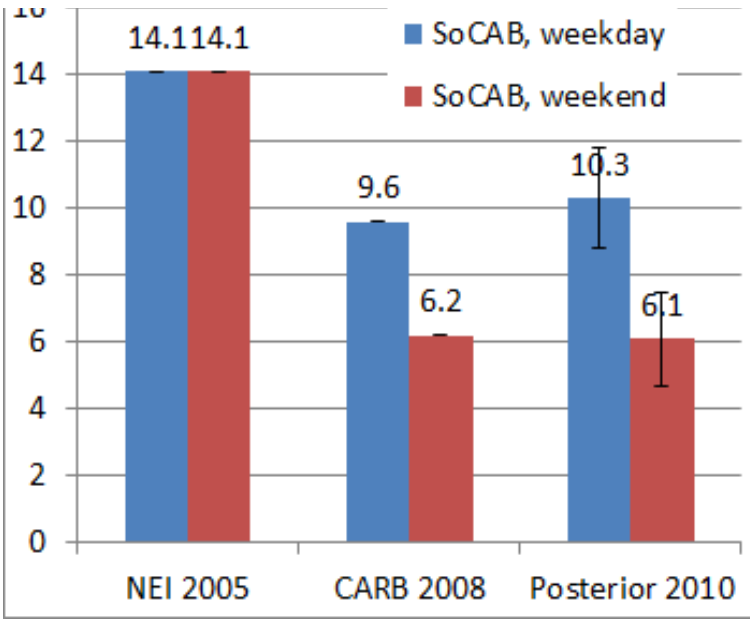
Inverse modeling applied to aircraft measurements helps to improve existing emission inventories at mesoscale

Example: Inverse Modeling of Urban NOx Emissions



- Aircraft observations lower inverse modeled emissions compared with NEI2005 prior
- Observations show weekend-weekday variations

Bottom-up & Posterior NOx Flux, kg/s



- CARB inventory captures emissions magnitude & day-of-week cycle better than NEI

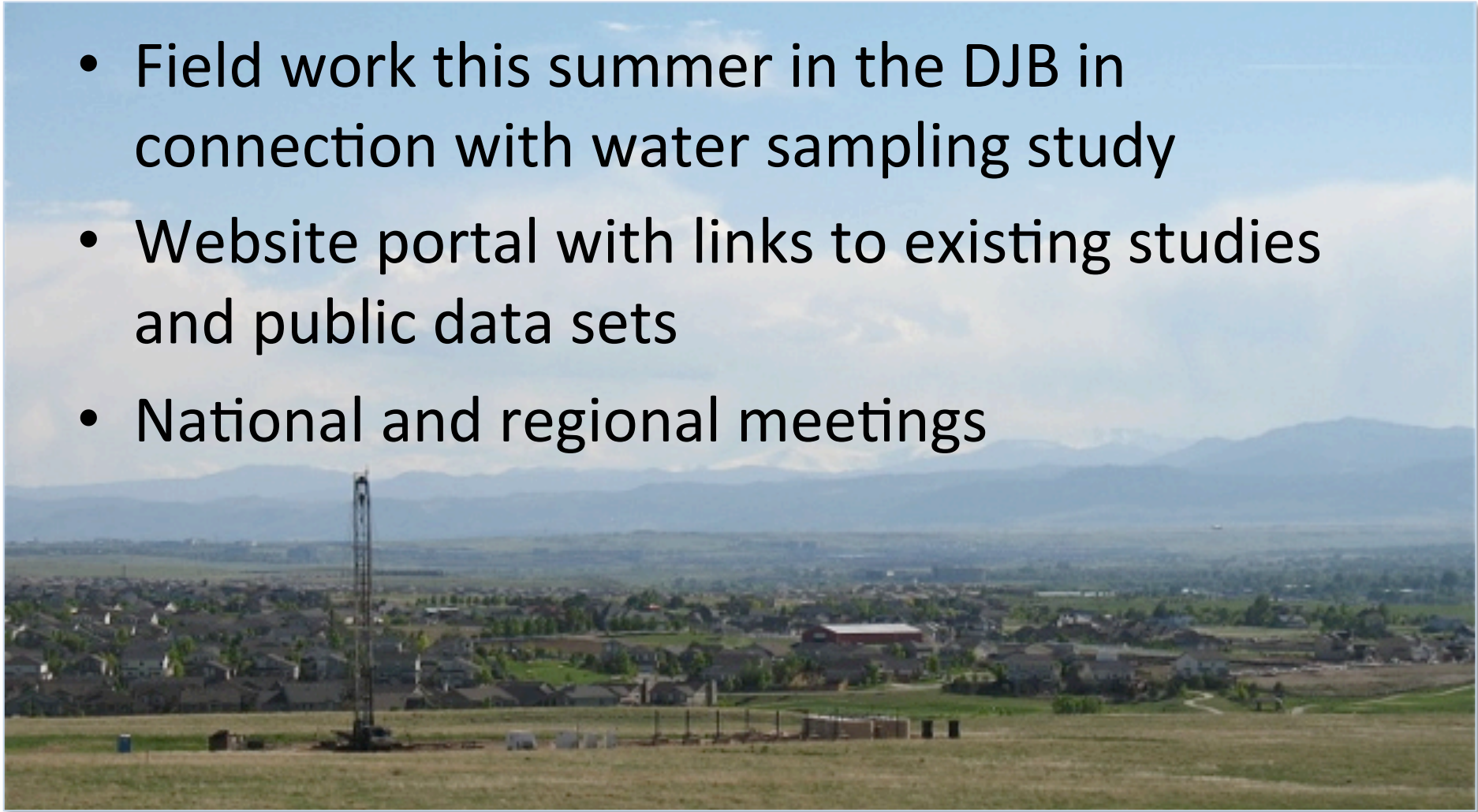
Concluding remarks

- ❑ NOAA's atmospheric measurements can be used to quantitatively assess emissions from oil and gas upstream and midstream activities
 - Our top-down emission estimates are
 - for a specific location and time
 - integrated fluxes from various O&G operations
- ❑ These types of studies provide objective evaluations of bottom-up inventories
 - Specifically can be used to assess at the regional scale
 - new inventory methodologies
 - impact of new regulations/practices
- ❑ NOAA measurements and modeling probe the processes that lead to air quality and climate impacts
 - We examine ozone distributions, chemical reactivity, basin/boundary layer dynamics, long-range transport, etc.
- ❑ Results from several on-going analyses should be available later this year



Planned work for CU NSF SRN

- Field work this summer in the DJB in connection with water sampling study
- Website portal with links to existing studies and public data sets
- National and regional meetings



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